

#### IN THE UNITED STATES PATENT OFFICE

I, Stephen DRANE BSc PhD BDÜ,

translator to RWS Translations Ltd., of Europa House, Marsham Way, Gerrards Cross, Buckinghamshire, England declare;

- 1. That I am a citizen of the United Kingdom of Great Britain and Northern Ireland.
- 2. That I am well acquainted with the German and English languages.
- 3. That the attached is, to the best of my knowledge and belief, a true translation into the English language of the accompanying copy of the specification filed with the application for a patent in Germany on 7 September 1995 under the number 195 33 023.4 and the official certificate attached hereto.
- 4. That I believe that all statements made herein of my own knowledge are true and that all statements made on information and belief are true; and further that these statements are made with the knowledge that wilful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such wilful false statements may jeopardise the validity of the patent application in the United States of America or any patent issuing thereon.

For and on behalf of RWS Translations Ltd.

The 26th day of February 1998

# FEDERAL REPUBLIC OF GERMANY CERTIFICATE

# BASF Aktiengesellschaft

of

### 67056 Ludwigshafen

have filed a Patent Application under the title:

"Novel carboxylic acid derivatives, their preparation and use"

on 7 September 1995 at the German Patent Office, and declare that they claim, for this application, the inner priority of the Application made in the Federal Republic of Germany on 14 October 1994, File No. P 44 36 851.8.

The attached document is a correct and accurate reproduction of the original submission for this Patent Application.

The German Patent Office has for the time being given the Application the symbols C 07 D 239/60, C 07 D 239/70, C 07 D 403/12, C 07 D 405/12, C 07 D 491/044, C 07 D 487/04, C 07 D 495/04, C 07 D 409/12, C 07 D 413/12, C 07 D 417/12, A 61 K 31/41 and A 61 K 31/505 of the International Patent Classification.

Munich, 16 October 1995

President of the German Patent Office

pp

Nietiedt

File No: 195 33 023.4

We claim:

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A carboxylic acid derivative of the formula I

where R is formyl, tetrazole [sic], nitrile [sic], a COOH group 15 or a radical which can be hydrolyzed to COOH, and the other substituents have the following meanings:

- R<sup>2</sup> hydrogen, hydroxyl, NH<sub>2</sub>, NH(C<sub>1</sub>-C<sub>4</sub>-alkyl), N(C<sub>1</sub>-C<sub>4</sub>-alkyl)<sub>2</sub>,
  halogen, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy,
  C<sub>1</sub>-C<sub>4</sub>-haloalkoxy or C<sub>1</sub>-C<sub>4</sub>-alkylthio;
  - nitrogen or  $CR^{14}$  where  $R^{14}$  is hydrogen or  $C_{1-5}$ -alkyl, or  $CR^{14}$  forms together with  $CR^3$  a 5- or 6-membered alkylene or alkenylene ring which can be substituted by one or two  $C_{1-4}$ -alkyl groups and in which in each case a methylene group can be replaced by oxygen, sulfur, -NH or -NC<sub>1-4</sub>-alkyl;
- R<sup>3</sup> hydrogen, hydroxyl, NH<sub>2</sub>, NH(C<sub>1</sub>-C<sub>4</sub>-Alkyl), N(C<sub>1</sub>-C<sub>4</sub>-alkyl)<sub>2</sub>, halogen, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy,
  C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, -NH-O-C<sub>1-4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-alkylthio or CR<sup>3</sup> is linked to CR<sup>14</sup> as indicated above to give a 5- or 6-membered ring;
  - R<sup>4</sup> and R<sup>5</sup> (which can be identical or different):

phenyl or naphthyl, which can be substituted by one or more of the following radicals: halogen, nitro, cyano, hydroxyl, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, phenoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, amino, C<sub>1</sub>-C<sub>4</sub>-alkylamino or C<sub>1</sub>-C<sub>4</sub>-dialkylamino; or

phenyl or naphthyl, which are connected together in the ortho positions via a direct linkage, a methylene, ethylene or ethenylene group, an oxygen or sulfur atom or an SO<sub>2</sub>, NH or N-alkyl group;

or C<sub>3</sub>-C<sub>7</sub>-cycloalkyl;

 $R^6$ hydrogen,  $C_1$ - $C_8$ -alkyl,  $C_3$ - $C_6$ -alkenyl,  $C_3$ - $C_6$ -alkynyl or C<sub>3</sub>-C<sub>8</sub>-cycloalkyl, where each of these radicals can be substituted one or more times by: halogen, nitro, cyano,  $C_1-C_4$ -alkoxy,  $C_3-C_6$ -alkenyloxy,  $C_3-C_6$ -alkynyloxy,  $C_1-C_4$ -alkylthio, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylcarbonyl, C<sub>1</sub>-C<sub>4</sub>-alkoxycarbonyl,  $C_{3-8}$ -alkylcarbonylalkyl,  $C_1$ - $C_4$ -alkylamino,  $di-C_1-C_4-alkylamino$ , phenyl, or phenyl or phenoxy which is substituted one or more times, eg. one to three times, by halogen, nitro, cyano, C1-C4-alkyl, C1-C4-haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy or  $C_1-C_4$ -alkylthio; 10

> phenyl or naphthyl, each of which can be substituted by one or more of the following radicals: halogen, nitro, cyano, hydroxyl, amino,  $C_1-C_4$ -alkyl,  $C_1-C_4$ -haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy, phenoxy,  $C_1-C_4$ -alkylthio,  $C_1-C_4$ -alkylamino, C<sub>1</sub>-C<sub>4</sub>-dialkylamino, dioxomethylene [sic] or dioxoethylene [sic];

a five- or six-membered heteroaromatic moiety containing one 20 to three nitrogen atoms and/or one sulfur or oxygen atom, which can carry one to four halogen atoms and/or one or two of the following radicals: C1-C4-alkyl, C1-C4-haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy,  $C_1-C_4$ -alkylthio, phenyl, phenoxy or phenylcarbonyl, it being possible for the phenyl 25 radicals in turn to carry one to five halogen atoms and/or one to three of the following radicals: C1-C4-alkyl,  $C_1-C_4$ -haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy and/or  $C_1-C_4$ -alkylthio;

30 with the proviso that  $R^6$  can be hydrogen only when Z is not a single bond;

- Y sulfur or oxygen or a single bond;
- 35 Z sulfur, oxygen, -SO-, -SO<sub>2</sub>- or a single bond.

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Novel carboxylic acid derivatives, their preparation and use

The present invention relates to novel carboxylic acid deriva-5 tives, their preparation and use.

Endothelin is a peptide which is composed of 21 amino acids and is synthesized and released by the vascular endothelium. Endothelin exists in three isoforms, ET-1, ET-2 and ET-3. In the following text, "endothelin" or "ET" signifies one or all isoforms of endothelin. Endothelin is a potent vasoconstrictor and has a potent effect on vessel tone. It is known that this vasoconstriction is caused by binding of endothelin to its receptor (Nature, 332, (1988) 411-415; FEBS Letters, 231, (1988) 440-444 and Biochem. Biophys. Res. Commun., 154, (1988) 868-875).

Increased or abnormal release of endothelin causes persistent vasoconstriction in the peripheral, renal and cerebral blood vessels, which may lead to illnesses. It has been reported in the literature that elevated plasma levels of endothelin were found in patients with hypertension, acute myocardial infarct, pulmonary hypertension, Raynaud's syndrome, atherosclerosis and in the airways of asthmatics (Japan J. Hypertension, 12, (1989) 79, J. Vascular Med. Biology 2, (1990) 207, J. Am. Med. Association 25 264, (1990) 2868).

Accordingly, substances which specifically inhibit the binding of endothelin to the receptor ought also to antagonize the various abovementioned physiological effects of endothelin and therefore 30 be valuable drugs.

We have found that certain carboxylic acid derivatives are good inhibitors of endothelin receptors.

35 The invention relates to carboxylic acid derivatives of the formula I

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$$R^{6} = Z - CH - Y - X$$

$$R^{5} = R$$

$$R^{3}$$

$$R^{3}$$

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where R is formyl, tetrazolye [sic], nitrile [sic], a COOH group or a radical which can be hydrolyzed to COOH, and the other substituents have the following meanings:

- 5 R<sup>2</sup> hydrogen, hydroxyl, NH<sub>2</sub>, NH( $C_1$ - $C_4$ -alkyl), N( $C_1$ - $C_4$ -alkyl)<sub>2</sub>, halogen,  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -haloalkoxy or  $C_1$ - $C_4$ -alkylthio;
- Notation of CR<sup>14</sup> where R<sup>14</sup> is hydrogen of C<sub>1-5</sub>-alkyl, of CR<sup>14</sup> forms together with CR<sup>3</sup> a 5- or 6-membered alkylene or alkenylene ring which can be substituted by one of two C<sub>1-4</sub>-alkyl groups and in which in each case a methylene group can be replaced by oxygen, sulfur, -NH or -NC<sub>1-4</sub>-alkyl;
- 15 R³ hydrogen, hydroxyl, NH<sub>2</sub>, NH(C<sub>1</sub>-C<sub>4</sub>-Alkyl), N(C<sub>1</sub>-C<sub>4</sub>-alkyl)<sub>2</sub>, halogen, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, -NH-O-C<sub>1-4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-alkylthio or CR³ is linked to CR¹⁴ as indicated above to give a 5- or 6-membered ring;
  - R4 and R5 (which can be identical or different):
- phenyl or naphthyl, which can be substituted by one or more of the following radicals: halogen, nitro, cyano, hydroxyl, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, phenoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, amino, C<sub>1</sub>-C<sub>4</sub>-alkylamino or C<sub>1</sub>-C<sub>4</sub>-dialkylamino; or
- phenyl or naphthyl, which are connected together in the ortho positions via a direct linkage, a methylene, ethylene or ethenylene group, an oxygen or sulfur atom or an SO<sub>2</sub>-, NH- or N-alkyl group, or C<sub>3</sub>-C<sub>7</sub>-cycloalkyl;
- R<sup>6</sup> hydrogen, C<sub>1</sub>-C<sub>8</sub>-alkyl, C<sub>3</sub>-C<sub>6</sub>-alkenyl, C<sub>3</sub>-C<sub>6</sub>-alkynyl or C<sub>3</sub>-C<sub>8</sub>-cycloalkyl, where each of these radicals can be substituted one or more times by: halogen, nitro, cyano, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>3</sub>-C<sub>6</sub>-alkenyloxy, C<sub>3</sub>-C<sub>6</sub>-alkynyloxy, C<sub>1</sub>-C<sub>4</sub>-alkyltio, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylcarbonyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy-carbonyl, C<sub>3-8</sub>-alkylcarbonylalkyl, C<sub>1</sub>-C<sub>4</sub>-alkylamino, di-C<sub>1</sub>-C<sub>4</sub>-alkylamino, phenyl, or phenyl or phenoxy which is
- substituted one or more times, eg. one to three times, by halogen, mitro, cyano,  $C_1-C_4$ -alkyl,  $C_1-C_4$ -haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy or  $C_1-C_4$ -alkylthio;
- phenyl or naphthyl, each of which can be substituted by one or more of the following radicals: halogen, nitro, cyano, hydroxyl, amino, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy,

 $C_1$ - $C_4$ -haloalkoxy, phenoxy,  $C_1$ - $C_4$ -alkylthio,  $C_1$ - $C_4$ -alkylamino,  $C_1$ - $C_4$ -dialkylamino, dioxomethylene [sic] or dioxoethylene [sic];

- a five- or six-membered heteroaromatic moiety containing one to three nitrogen atoms and/or one sulfur or oxygen atom, which can carry one to four halogen atoms and/or one or two of the following radicals:  $C_1-C_4$ -alkyl,  $C_1-C_4$ -haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy,  $C_1-C_4$ -alkylthio, phenyl,
- phenoxy or phenylcarbonyl, it being possible for the phenyl radicals in turn to carry one to five halogen atoms and/or one to three of the following radicals: C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy and/or C<sub>1</sub>-C<sub>4</sub>-alkylthio;

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with the proviso that  $R^6$  can be hydrogen only when Z is not a single bond;

Y sulfur or oxygen or a single bond;

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z sulfur or oxygen or a single bond.

The compounds, and the intermediates for preparing them, such as IV and VI, may have one or more asymmetrical [sic] substituted

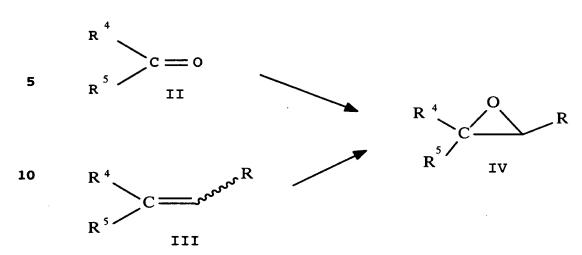
25 carbon atoms. Such compounds may be in the form of the pure enantiomers or pure diastereomers or a mixture thereof. The use of an enant-iomerically pure compound as active substance is preferred.

The invention furthermore relates to the use of the abovemen-30 tioned carboxylic acid derivatives for producing drugs, in particular for producing endothelin receptor inhibitors.

The invention furthermore relates to the preparation of the compounds of the formula IV in enantiomerically pure form. Enantio-35 selective epoxidation of an olefin with two phenyl substituents is known (J. Org. Chem. <u>59</u>, 1994, 4378-4380). We have now found, surprisingly, that even ester groups in these systems permit epoxidation in high optical purity.

40 The preparation of the compounds according to the invention where Z is sulfur or oxygen starts from the epoxides IV, which are obtained in a conventional manner, eg. as described in J. March, Advanced Organic Chemistry, 2nd ed., 1983, page 862 and page 750, from the ketones II or the olefins III:

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15 Carboxylic acid derivatives of the general formula VI can be prepared by reacting the epoxides of the general formula IV (eg. with  $R = ROOR^{10}$  [sic]) with alcohols or thiols of the general formula V where  $R^6$  and Z have the meanings stated in claim 1.

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IV + R<sup>6</sup>ZH  $\longrightarrow$  R  $\stackrel{6}{\longrightarrow}$  Z  $\stackrel{}{\longrightarrow}$  CH  $\longrightarrow$  OH VI

V

To do this, compounds of the general formula IV are heated with compounds of the formula V, in the molar ratio of about 1:1 to 1:7, preferably 1 to 3 mole equivalents, to 50-200°C, preferably 80-150°C.

The reaction can also take place in the presence of a diluent. All solvents which are inert toward the reagents used can be used for this purpose.

35 Examples of such solvents or diluents are water, aliphatic, alicyclic and aromatic hydrocarbons, which may in each case be chlorinated, such as hexane, cyclohexane, petroleum ether, naphtha, benzene, toluene, xylene, methylene chloride, chloroform, carbon tetrachloride, ethyl chloride and trichloroethylene, ethers such as diisopropyl ether, dibutyl ether, methyl tert-butyl ether,

40 as diisopropyl ether, dibutyl ether, methyl tert-butyl ether, propylene oxide, dioxane and tetrahydrofuran, ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone and methyl isobutyl ketone, nitriles such as acetonitrile and propionitrile, alcohols, such as methanol, ethanol, isopropanol, butanol and

45 ethylene glycol, esters such as ethyl acetate and amyl acetate, amides such as dimethylformamide, dimethylacetamide and N-methylpyrrolidone, sulfoxides and sulfones, such as dimethyl sulfoxide

and sulfolane, bases such as pyridine, cyclic ureas such as 1,3-dimethylimidazolidin-2-one and 1,3-dimethyl-3,4,5,6-tetra-hydro-2(1H)-pyrimidinone.

5 The reaction is preferably carried out at a temperature in the range from 0°C to the boiling point of the solvent or mixture of solvents.

The presence of a catalyst may be advantageous. Suitable cata10 lysts are strong organic and inorganic acids, and Lewis acids.

Examples thereof are, inter alia, sulfuric acid, hydrochloric acid, trifluoroacetic acid, p-toluenesulfonic acid, boron trifluoride etherate and titanium(IV) alcoholates.

- 15 Compounds of the formula VI where  $R^4$  and  $R^5$  are cycloalkyl can also be prepared by subjecting compounds of the formula VI where  $R^4$  and  $R^5$  are phenyl, naphthyl, or phenyl or naphthyl substituted as described above, to a nuclear hydrogenation.
- 20 Compounds of the formula VI can be obtained in enantiomerically pure form by starting from enantiomerically pure compounds of the formula IV and reacting them in the manner described with compounds of the formula V.
- 25 It is furthermore possible to obtain enantiomerically pure compounds of the formula VI by carrying out a classical racemate resolution on racemic or diastereomeric compounds of the formula VI using suitable enantiomerically pure bases such as brucine, strychnine, quinine, quinidine, chinchonidine [sic], chinchonine
- 30 [sic], yohimbine, morphine, dehydroabietylamine, ephedrine (-),
   (+), deoxyephedrine (+), (-), threo-2-amino-1-(p-nitrophe nyl)-1,3-propanediol (+), (-), threo-2-(N,N-dimethylamino)-1-(p nitrophenyl)-1,3-propanediol (+), (-) threo-2-amino-1-phenyl 1,3-propanediol (+), (-), α-methylbenzylamine (+), (-),
- 35 α-(1-naphthyl)ethylamine (+), (-), α-(2-naphthyl)ethylamine (+),
   (-), aminomethylpinane, N,N-dimethyl-1-phenylethylamine,
   N-methyl-1-phenylethylamine, 4-nitrophenylethylamine,
   pseudoephedrine, norephedrine, norpseudoephedrine, amino acid
   derivatives, peptide derivatives.

The compounds according to the invention where Y is oxygen, and the remaining substituents have the meanings stated under the general formula I, can be prepared, for example, by reacting the carboxylic acid derivatives of the general formula VI where the

45 substituents have the stated meanings with compounds of the general formula VII

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where  $R^{15}$  is halogen or  $R^{16}$ -SO<sub>2</sub>-, where  $R^{16}$  can be  $C_1$ - $C_4$ -alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl or phenyl. The reaction preferably takes place in one of the abovementioned inert diluents with the addition of a suitable base, ie. of a base which deprotonates the intermediate 15 VI, in a temperature range from room temperature to the boiling point of the solvent.

Compounds of the formula VII are known, some of them can be bought, or they can be prepared in a generally known manner.

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It is possible to use as base an alkali metal or alkaline earth metal hydride such as sodium hydride, potassium hydride or calcium hydride, a carbonate such as an alkali metal carbonate, eg. sodium or potassium carbonate, an alkali metal or alkaline earth 25 metal hydroxide such as sodium or potassium hydroxide, an organometallic compound such as butyllithium, or an alkali metal amide such as lithium diisopropylamide.

The compounds according to the invention where Y is sulfur, and 30 the remaining substituents have the meanings stated under the general formula I, can be prepared, for example, by reacting carboxylic acid derivatives of the general formula VIII, which can be obtained in a known manner from compounds of the general formula VI and in which the substituents have the abovementioned 35 meanings, with compounds of the general formula IX, where  $R^2$ ,  $R^3$ and X have the meanings stated under general formula I.

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The reaction preferably takes place in one of the abovementioned inert diluents with the addition of a suitable base, ie. a base which deprotonates the intermediate IX, in a temperature range from room temperature to the boiling point of the solvent.

It is possible to use as base, besides those mentioned above, organic bases such as triethylamine, pyridine, imidazole or diazabicycloundecane [sic].

10 Carboxylic acid derivatives of the formula VIa (Z in formula VI =
 direct linkage) can be prepared by reacting epoxides of the for mula IV with cuprates of the formula XI:

15
$$IV + R_{2}^{6} Cu(CN)Li_{2} \longrightarrow R - C - CH - OH$$

$$R = R + C - CH - OH$$

$$R = R + C - CH - OH$$

$$R = R + C - CH - OH$$

$$R = R + C - CH - OH$$

$$R = R + C - CH - OH$$

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$$R = R + C - CH - OH$$

$$R = R + C - CH - CH$$

$$R = R + C - CH$$

$$R = R + CH$$

$$R = R$$

The cuprates can be prepared as described in Tetrahedron Letters 23, (1982) 3755.

Compounds of the formula I can also be prepared by starting from the corresponding carboxylic acids, ie. compounds of the formula I where R is COOH, and initially converting these in a conventional manner into an activated form, such as a halide, an anhydride or imidazolide, and then reacting the latter with an appropriate hydroxy compound HOR10. This reaction can be carried out in the usual solvents and often requires addition of a base, in which case those mentioned above are suitable. These two steps can also be simplified, for example, by allowing the carboxylic acid to act on the hydroxy compound in the presence of a dehydrating agent such as a carbodiimide.

In addition, it is also possible for compounds of the formula I to be prepared by starting from the salts of the corresponding 40 carboxylic acids, ie. from compounds of the formula I where R is COR<sup>1</sup> and R<sup>1</sup> is OM, where M can be an alkali metal cation or the equivalent of an alkaline earth metal cation. These salts can be reacted with many compounds of the formula R<sup>1</sup>-A where A is a conventional nucleofugic leaving group, for example halogen such as 45 chlorine, bromine, iodine or aryl- or alkylsulfonyl which is unsubstituted or substituted by halogen, alkyl or haloalkyl, such as toluenesulfonyl and methylsulfonyl, or another equivalent

leaving group. Compounds of the formula R<sup>1</sup>-A with a reactive substituent A are known or can be easily obtained with general expert knowledge. This reaction can be carried out in conventional solvents and advantageously takes place with the addition of a 5 base, in which case those mentioned above are suitable.

The radical R in formula I may vary widely. For example, R is a group

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15 where R1 has the following meanings:

- a) hydrogen;
- b) succinylimidoxy [sic];

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- c) a five-membered heteroaromatic moiety linked by a nitrogen atom, such as pyrrolyl, pyrazolyl, imidazolyl and triazolyl, which may carry one or two halogen atoms, in particular fluorine and chlorine and/or one or two of the following radicals:
  - C<sub>1</sub>-C<sub>4</sub>-alkyl such as methyl, ethyl, 1-propyl, 2-propyl, 2-methyl-2-propyl, 2-methyl-1-propyl, 1-butyl, 2-butyl;
- C<sub>1</sub>-C<sub>4</sub>-haloalkyl, in particular C<sub>1</sub>-C<sub>2</sub>-haloalkyl such as fluoromethyl, difluoromethyl, trifluoromethyl, chlorodifluoromethyl, dichlorofluoromethyl, trichloromethyl, 1-fluoroethyl, 2-fluoroethyl, 2,2-difluoroethyl, 2,2,2-trifluoroethyl, 2-chloro-2,2-difluoroethyl, 2,2-dichlorodifluoroethyl, 2,2,2-trichloroethyl, 2,2,2-trichloroethyl, 2,2,2-trichloroethyl, 2,2,2-trichloroethyl, 2,2,2-trichloroethyl, 2,2,2-trichloroethyl, 2,2,2-trichloroethyl,

C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, in particular C<sub>1</sub>-C<sub>2</sub>-haloalkoxy such as difluoromethoxy, trifluoromethoxy, chlorodifluoromethoxy, 1-fluoroethoxy, 2-fluoroethoxy, 2,2-difluoroethoxy, 1,1,2,2-tetrafluoroethoxy, 2,2,2-trifluoroethoxy, 2-chloro-1,1,2-trifluoroethoxy and pentafluoroethoxy, in particular trifluoromethoxy;

C<sub>1</sub>-C<sub>4</sub>-alkoxy such as methoxy, ethoxy, propoxy, 1-methylethoxy, butoxy, 1-methylpropoxy, 2-methylpropoxy, 1,1-dimethylethoxy, in particular methoxy, ethoxy, 1-methylethoxy;

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 $C_1$ - $C_4$ -alkylthio such as methylthio, ethylthio, propylthio, 1-methylethylthio, butylthio, 1-methylpropylthio, 2-methylpropylthio, 1,1-dimethylethylthio, in particular methylthio and ethylthio;

d) R1 furthermore a radical

3-methyl-2-pentenyl;

 $-(0)_{m} N \setminus_{\mathbb{R}^{8}}^{\mathbb{R}^{7}}$ 

where m is 0 or 1 and  $R^7$  and  $R^8$ , which can be identical or different, have the following meanings:

hydrogen

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C<sub>1</sub>-C<sub>8</sub>-alkyl, in particular C<sub>1</sub>-C<sub>4</sub>-alkyl as mentioned above;

20 C<sub>3</sub>-C<sub>6</sub>-alkenyl such as 2-propenyl, 2-butenyl, 3-butenyl, 1-methyl-2-propenyl, 2-methyl-2-propenyl, 2-pentenyl, 3-pentenyl, 4-pentenyl, 1-methyl-2-butenyl, 2-methyl-2-butenyl, 3-methyl-2-butenyl, 1-methyl-3-butenyl, 2-methyl-3-butenyl, 3-methyl-3-butenyl, 1,1-dimethyl-25 2-propenyl, 1,2-dimethyl-2-propenyl, 1-ethyl-2-propenyl, 2-hexenyl, 3-hexenyl, 4-hexenyl, 5-hexenyl, 1-methyl-2-pentenyl, 2-methyl-2-pentenyl, 3-methyl-2-pentenyl, 4-methyl-2-pentenyl, 3-methyl-3-pentenyl, 4-methyl-3-pentenyl, 1-methyl-4-pentenyl, 2-methyl-4-pentenyl, 3-methyl-4-pent-30 enyl, 4-methyl-4-pentenyl, 1,1-dimethyl-2-butenyl, 1,1-dimethyl-3-butenyl, 1,2-dimethyl-2-butenyl, 1,2-dimethyl-3-butenyl, 1,3-dimethyl-2-butenyl, 1,3-dimethyl-3-butenyl, 2,2-dimethyl-3-butenyl, 2,3-dimethyl-2-butenyl, 2,3-dimethyl-3-butenyl, 1-ethyl-2-butenyl, 1-ethyl-3-butenyl, 2-ethyl-35 2-butenyl, 2-ethyl-3-butenyl, 1,1,2-trimethyl-2-propenyl, 1-ethyl-1-methyl-2-propenyl and 1-ethyl-2-methyl-2-propenyl, in particular 2-propenyl, 2-butenyl, 3-methyl-2-butenyl and

C<sub>3</sub>-C<sub>6</sub>-alkynyl such as 2-propynyl, 2-butynyl, 3-butynyl,
1-methyl-2-propynyl, 2-pentynyl, 3-pentynyl, 4-pentynyl,
1-methyl-3-butynyl, 2-methyl-3-butynyl, 1-methyl-2-butynyl,
1,1-dimethyl-2-propynyl, 1-ethyl-2-propynyl, 2-hexynyl,
3-hexynyl, 4-hexynyl, 5-hexynyl, 1-methyl-2-pentynyl,
1-methyl-2-pentynyl, 1-methyl-3-pentynyl,
1-methyl-4-pentynyl, 2-methyl-3-pentynyl,
2-methyl-4-pentynyl, 3-methyl-4-pentynyl,

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4-methyl-2-pentynyl, 1,1-dimethyl-2-butynyl,
1,1-dimethyl-3-butynyl, 1,2-dimethyl-3-butynyl,
2,2-dimethyl-3-butynyl, 1-ethyl-2-butynyl, 1-ethyl-3-butynyl,
2-ethyl-3-butynyl and 1-ethyl-1-methyl-2-propynyl, preferably
2-propynyl, 2-butynyl, 1-methyl-2-propynyl and
1-methyl-2-butynyl, in particular 2-propynyl

C<sub>3</sub>-C<sub>8</sub>-cycloalkyl such as cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl, cyclooctyl, where these alkyl, cycloalkyl, alkenyl and alkynyl groups can each carry one to five halogen atoms, in particular fluorine or chlorine and/or one or two of the following groups:

- C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy as mentioned above, C<sub>3</sub>-C<sub>6</sub>-alkenyloxy, C<sub>3</sub>-C<sub>6</sub>-alkenylthio, C<sub>3</sub>-C<sub>6</sub>-alkynyloxy, C<sub>3</sub>-C<sub>6</sub>-alkynylthio, where the alkenyl and alkynyl constituents present in these radicals preferably have the abovementioned meanings;
- C<sub>1</sub>-C<sub>4</sub>-alkylcarbonyl such as, in particular, methylcarbonyl, ethylcarbonyl, propylcarbonyl, 1-methylethylcarbonyl, butylcarbonyl, 1-methylpropylcarbonyl, 2-methylpropylcarbonyl, 1,1-dimethylethylcarbonyl;
- C<sub>1</sub>-C<sub>4</sub>-alkoxycarbonyl such as methoxycarbonyl, ethoxycarbonyl, propyloxycarbonyl, 1-methylethoxycarbonyl, butyloxycarbonyl, 1-methylpropyloxycarbonyl, 2-methylpropyloxycarbonyl, 1,1-dimethylethoxycarbonyl;
- 30  $C_3-C_6$ -alkenylcarbonyl,  $C_3-C_6$ -alkynylcarbonyl,  $C_3-C_6$ -alkenyloxy-carbonyl and  $C_3-C_6$ -alkynyloxycarbonyl, where the alkenyl and alkynyl radicals are preferably defined as detailed above;
- phenyl, unsubstituted or substituted one or more times, eg.
  one to three times, by halogen, nitro, cyano, C<sub>1</sub>-C<sub>4</sub>-alkyl,
  C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy or C<sub>1</sub>-C<sub>4</sub>-alkylthio, such as 2-fluorophenyl, 3-chlorophenyl, 4-bromophenyl,
  2-methylphenyl, 3-nitrophenyl, 4-cyanophenyl, 2-trifluoromethylphenyl, 3-methoxyphenyl, 4-trifluoroethoxyphenyl,
- 2-methylthiophenyl, 2,4-dichlorophenyl, 2-methoxy-3-methyl-phenyl, 2,4-dimethoxyphenyl, 2-nitro-5-cyanophenyl, 2,6-difluorophenyl;

di-C<sub>1</sub>-C<sub>4</sub>-alkylamino such as, in particular, dimethylamino, dipropylamino, N-propyl-N-methylamino, N-propyl-N-ethylamino, diisopropylamino, N-isopropyl-N-methylamino, N-isopropyl-N-propylamino;

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 $R^7$  and  $R^8$  furthermore phenyl which can be substituted by one or more, eg. one to three, of the following radicals: halogen, nitro, cyano,  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -haloalkoxy or  $C_1$ - $C_4$ -alkylthio, as mentioned above in particular;

or  $R^7$  and  $R^8$  together form a  $C_4$ - $C_7$ -alkylene chain which is closed to form a ring, is unsubstituted or substituted, eg. substituted by  $C_1$ - $C_4$ -alkyl, and may contain a heteroatom selected from the group consisting of oxygen, sulfur or nitrogen, such as  $-(CH_2)_4$ -,  $-(CH_2)_5$ -,  $-(CH_2)_6$ -,  $-(CH_2)_7$ -,  $-(CH_2)_2$ -O- $(CH_2)_2$ -,  $-(CH_2)_3$ -,  $-(CH_2)_2$ -O- $(CH_2)_3$ -,  $-(CH_2)_3$ -,

20 e) R1 furthermore a group

$$-0-(CH_2)_p-S-R^9$$

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where k is 0, 1 and 2, p is 1, 2, 3 and 4 and  $R^9$  is

 $C_1-C_4$ -alkyl,  $C_1-C_4$ -haloalkyl,  $C_3-C_6$ -alkenyl,  $C_3-C_6$ -alkynyl or unsubstituted or substituted phenyl, as mentioned above in particular.

f) R1 furthermore a radical OR10, where R10 is:

hydrogen, the cation of an alkali metal such as lithium, sodium, potassium or the cation of an alkaline earth metal such as calcium, magnesium and barium or an environmentally compatible organic ammonium ion such as tertiary  $C_1-C_4$ -alkylammonium or the ammonium ion;

40  $C_3-C_8$ -cycloalkyl as mentioned above, which may carry one to three  $C_1-C_4$ -alkyl groups;

C<sub>1</sub>-C<sub>8</sub>-alkyl such as, in particular, methyl, ethyl, propyl, 1-methylethyl, butyl, 1-methylpropyl, 2-methylpropyl, 1,1-dimethylethyl, pentyl, 1-methylbutyl, 2-methylbutyl, 3-methylbutyl, 1,2-dimethylpropyl, 1,1-dimethylpropyl, 2,2-dimethylpropyl, 1-ethylpropyl, hexyl, 1-methylpentyl,

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2-methylpentyl, 3-methylpentyl, 4-methylpentyl, 1,2-dimethylbutyl, 1,3-dimethylbutyl, 2,3-dimethylbutyl, 1,1-dimethylbutyl, 2,2-dimethylbutyl, 3,3-dimethylbutyl, 1,1,2-trimethylpropyl, 1,2,2-trimethylpropyl, 1-ethylbutyl, 2-ethylbutyl, 1-ethyl-2-methylpropyl, which can carry one to five halogen atoms, in particular fluorine and chlorine and/or one of the following radicals:

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 $C_1-C_4$ -alkoxy,  $C_1-C_4$ -alkylthio, cyano,  $C_1-C_4$ -alkylcarbonyl, 10  $C_3-C_8$ -cycloalkyl,  $C_1-C_4$ -alkoxycarbonyl, phenyl, phenoxy or phenylcarbonyl, where the aromatic radicals in turn can carry in each case one to five halogen atoms and/or one to three of the following radicals: nitro, cyano, C1-C4-alkyl, C1-C4-haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy and/or  $C_1-C_4$ -alkylthio, 15 as mentioned above in particular;

a C1-C8-alkyl as mentioned above, which can carry one to five halogen atoms, in particular fluorine and/or chlorine, and carries one of the following radicals: a 5-membered heteroaromatic moiety containing one to three nitrogen atoms, or a 5-membered heteroaromatic moiety containing a nitrogen atom and an oxygen or sulfur atom, which can carry one to four halogen atoms and/or one or two of the following radicals:

nitro, cyano,  $C_1-C_4$ -alkyl,  $C_1-C_4$ -haloalkyl,  $C_1-C_4$ -alkoxy, phenyl,  $C_1-C_4$ -haloalkoxy and/or  $C_1-C_4$ -alkylthio. Particular mention may be made of: 1-pyrazolyl, 3-methyl-1-pyrazolyl, 4-methyl-1-pyrazolyl, 3,5-dimethyl-1-pyrazolyl, 3-phenyl-1-pyrazolyl, 4-phenyl-1-pyrazolyl,

4-chloro-1-pyrazolyl, 4-bromo-1-pyrazolyl, 1-imidazolyl, 1-benzimidazolyl, 1,2,4-triazol-1-yl, 3-methyl-1,2,4-triazol-1-yl, 5-methyl-1,2,4-triazol-1-yl, 1-benzotriazolyl, 3-isopropyl-5-isoxazolyl, 3-methyl-5-isoxazolyl, 2-oxazolyl, 2-thiazolyl, 2-imidazolyl, 3-ethyl-5-isoxazolyl, 3-phenyl-5-isoxazolyl, 3-tert-butyl-5-isoxazolyl;

a C<sub>2</sub>-C<sub>6</sub>-alkyl group which carries one of the following radicals in position 2: C<sub>1</sub>-C<sub>4</sub>-alkoxyimino, C<sub>3</sub>-C<sub>6</sub>-alkynyloxyimino, C<sub>3</sub>-C<sub>6</sub>-haloalkenyloxyimino or benzyloxyimino;

a  $C_3-C_6$ -alkenyl or  $C_3-C_6$ -alkynyl group, it being possible for these groups in turn to carry one to five halogen atoms;

45 R10 furthermore a phenyl radical which can carry one to five halogen atoms and/or one to three of the following radicals: nitro, cyano, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy,

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 $C_1-C_4$ -haloalkoxy and/or  $C_1-C_4$ -alkylthio, as mentioned above in particular;

a 5-membered heteroaromatic moiety which is linked via a nitrogen atom, contains one to three nitrogen atoms and can carry one or two halogen atoms and/or one or two of the following radicals: C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, phenyl, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy and/or C<sub>1</sub>-C<sub>4</sub>-alkylthio. Particular mention may be made of: 1-pyrazolyl, 3-methyl-1-pyrazolyl, 4-methyl-1-pyrazolyl, 3,5-dimethyl-1-pyrazolyl, 3-phenyl-1-pyrazolyl, 4-phenyl-1-pyrazolyl, 4-chloro-1-pyrazolyl, 4-phenyl-1-pyrazolyl, 1-imidazolyl, 1-benzimidazolyl, 1,2,4-triazol-1-yl, 3-methyl-1,2,4-triazol-1-yl, 5-methyl-1,2,4-triazol-1-yl, 1-benzotriazolyl, 3,4-dichloro-1-imidazolyl;

R<sup>10</sup> furthermore a group

$$-- N = C <_{R^{-12}}^{R^{-11}}$$

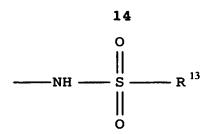
where  $R^{11}$  and  $R^{12}$ , which can be identical or different, are:

 $C_1$ - $C_8$ -alkyl,  $C_3$ - $C_6$ -alkenyl,  $C_3$ - $C_6$ -alkynyl,  $C_3$ - $C_8$ -cycloalkyl, it being possible for these radicals to carry a  $C_1$ - $C_4$ -alkylthio and/or an unsubstituted or substituted phenyl radical, as mentioned above in particular;

phenyl which can be substituted by one or more, eg. one to three, of the following radicals: halogen, nitro, cyano,  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -haloalkoxy or  $C_1$ - $C_4$ -alkylthio, where these radicals are, in particular, those mentioned above;

or  $R^{11}$  and  $R^{12}$  together form a  $C_3$ - $C_{12}$ -alkylene chain which can carry one to three  $C_1$ - $C_4$ -alkyl groups and contain a heteroatom from the group consisting of oxygen, sulfur and nitrogen, as mentioned in particular for  $R^7$  and  $R^8$ .

g) R<sup>1</sup> furthermore a radical

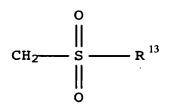


where R13 is:

- 10  $C_1-C_4$ -alkyl,  $C_3-C_6$ -alkenyl,  $C_3-C_6$ -alkynyl,  $C_3-C_8$ -cycloalkyl as mentioned above in particular, it being possible for these radicals to carry a  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -alkylthio and/or a phenyl radical as mentioned above;
- phenyl, unsubstituted or substituted, in particular as mentioned above.
  - h) R1 a radical

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where R13 has the abovementioned meaning.

- R can furthermore be:
- 30 tetrazole [sic] or nitrile [sic].

In respect of the biological effect, preferred carboxylic acid derivatives of the general formula I, both as pure enantiomers and pure diastereomers or as mixture thereof, are those where the 35 substituents have the following meanings:

- R<sup>2</sup> hydrogen, hydroxyl, N(C<sub>1</sub>-C<sub>4</sub>-alkyl)<sub>2</sub>, the C<sub>1</sub>-C<sub>4</sub>-alkyl,
  C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy,
  C<sub>1</sub>-C<sub>4</sub>-alkylthio groups and halogen atoms mentioned in detail
  for R<sup>1</sup>, especially chlorine, methyl, methoxy, ethoxy,
  difluoromethoxy, trifluoromethoxy;
- X nitrogen or CR14 where
- 45  $R^{14}$  is hydrogen or alkyl, or  $CR^{14}$  forms together with  $CR^3$  a 4- to 5-membered alkylene or alkenylene ring in which, in each case, a methylene group can be replaced by oxygen or sulfur,

such as  $-CH_2-CH_2-O-$ , -CH=CH-O-,  $-CH_2-CH_2-CH_2-O-$ ,  $-CH=CH-CH_2O-$ , in particular hydrogen,  $-CH_2-CH_2-O-$ ,  $-CH(CH_3)-CH(CH_3)-O-$ ,  $-C(CH_3)=C(CH_3)-O-$ ,  $-CH=C(CH_3)-O-$  or  $-C(CH_3)=C(CH_3)-S$ ;

- 5 R³ the hydrogen, hydroxyl, N(C<sub>1</sub>-C<sub>4</sub>-alkyl)<sub>2</sub>, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, C<sub>1</sub>-C<sub>4</sub>-alkyl-thio groups and halogen atoms mentioned for R¹, especially chlorine, methyl, methoxy, ethoxy, difluoromethoxy, trifluoromethoxy or is linked to R¹⁴ as mentioned above to give a 5- or 6-membered ring;
- R<sup>4</sup> and R<sup>5</sup> phenyl or naphthyl, which can be substituted by one or more, eg. one to three, of the following radicals: halogen, nitro, cyano, hydroxyl, mercapto, amino, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, C<sub>1</sub>-C<sub>4</sub>-alkylamino, di-C<sub>1</sub>-C<sub>4</sub>-alkylamino, C<sub>1</sub>-C<sub>4</sub>-alkylcarbonyl, C<sub>1</sub>-C<sub>4</sub>-alkoxycarbonyl; phenyl or naphthyl, which are connected together in the ortho positions by a direct linkage, a methylene, ethylene or ethenylene group, an oxygen or sulfur atom or an SO<sub>2</sub>, NH or N-alkyl group, or C<sub>3</sub>-C<sub>7</sub>-cycloalkyl;
- R<sup>6</sup> C<sub>1</sub>-C<sub>8</sub>-alkyl, C<sub>3</sub>-C<sub>6</sub>-alkenyl, C<sub>3</sub>-C<sub>6</sub>-alkynyl or C<sub>3</sub>-C<sub>8</sub>-cycloalkyl as mentioned above in particular, it being possible for these radicals in each case to be substituted one or more times by: halogen, hydroxyl, nitro, cyano, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>3</sub>-C<sub>6</sub>-alkenyl-oxy, C<sub>3</sub>-C<sub>6</sub>-alkynyloxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylcarbonyl, hydroxycarbonyl, C<sub>1</sub>-C<sub>4</sub>-alkoxycarbonyl, C<sub>1</sub>-C<sub>4</sub>-alkylamino, di-C<sub>1</sub>-C<sub>4</sub>-alkylamino or unsubstituted or substituted phenyl or phenoxy, as mentioned above in particular;

phenyl or naphthyl, which can be substituted by one or more of the following radicals: halogen, nitro, cyano, hydroxyl, amino, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, phenoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, C<sub>1</sub>-C<sub>4</sub>-akylamino [sic] or C<sub>1</sub>-C<sub>4</sub>-dialkylamino, as mentioned in particular for R<sup>7</sup> and R<sup>4</sup>;

a five- or six-membered heteroaromatic moiety which contains one to three nitrogen atoms and/or one sulfur or oxygen atom and which can carry one to four halogen atoms and/or one or two of the following radicals: C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-haloalkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-haloalkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio, phenyl, phenoxy or phenylcarbonyl, it being possible for the phenyl radicals in turn to carry one to five halogen atoms and/or one to three of the following radicals: C<sub>1</sub>-C<sub>4</sub>-alkyl,

 $C_1-C_4$ -haloalkyl,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -haloalkoxy and/or  $C_1-C_4$ -alkylthio, as mentioned for  $R^4$  in particular;

Y sulfur, oxygen or a single bond;

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z sulfur, oxygen, -SO-, -SO<sub>2</sub>- or a single bond.

Particularly preferred compounds of the formula I, both as pure enantiomers and pure diastereomers or as mixture thereof, are 10 those in which the substituents have the following meanings:

- $R^2$   $C_1-C_4$ -alkyl,  $C_1-C_4$ -alkoxy
- X nitrogen or CR14, where

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- is hydrogen or alkyl, or CR<sup>14</sup> forms together with CR<sup>3</sup> a 4- or 5-membered alkylene or alkenylene ring such as -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-, -CH=CH-CH<sub>2</sub>-, in which in each case a methylene group can be replaced by oxygen or sulfur, such as -CH<sub>2</sub>-CH<sub>2</sub>-O-, -CH=CH-O-, -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-O-, -CH=CH-CH<sub>2</sub>O-, in particular hydrogen,
- 20  $-CH_2-CH_2-O-$ ,  $-CH=CH-CH_2O-$ , in particular hydrogen,  $-CH_2-CH_2-O-$ ,  $-CH(CH_3)-CH(CH_3)-O-$ ,  $-C(CH_3)=C(CH_3)-O-$ ,  $-CH=C(CH_3)-O-$  or  $-C(CH_3)=C(CH_3)-S$ ;
- $R^3$  the  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -alkylthio groups 25 mentioned for  $R^1$ , or is linked to  $R^{14}$  as mentioned above to give a 5- or 6-membered ring;
- R<sup>4</sup> and R<sup>5</sup> phenyl (identical or different) which can be substituted by one or more, eg. one to three, of the following radicals: halogen, nitro, hydroxyl, C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio or
- R<sup>4</sup> and R<sup>5</sup> are phenyl groups which are connected together in the ortho positions by a direct linkage, a methylene, ethylene or ethenylene group, an oxygen or sulfur atom or an SO<sub>2</sub>, NH or N-alkyl group; or
  - $R^4$  and  $R^5$  are  $C_3-C_7$ -cycloalkyl;

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 $C_1-C_8$ -alkyl,  $C_3-C_6$ -alkenyl or  $C_3-C_8$ -cycloalkyl, it being possible for these radicals in each case to be substituted one or more times by: halogen, hydroxyl, nitro, cyano,  $C_1-C_4$ -alkoxy,  $C_3-C_6$ -alkenyloxy,  $C_1-C_4$ -alkylthio;

phenyl or naphthyl, which can be substituted by one or more of the following radicals: halogen, nitro, cyano, hydroxyl, amino,  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -akylamino [sic] or  $C_1$ - $C_4$ -dialkylamino;

a five- or six-membered heteroaromatic moiety which contains a nitrogen atom and/or a sulfur or oxygen atom and which can carry one to four halogen atoms and/or one or two of the following radicals:  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -alkylthio, phenyl, phenoxy or phenylcarbonyl, it being possible for the phenyl radicals in turn to carry one to five halogen atoms and/or one to three of the following radicals:  $C_1$ - $C_4$ -alkyl,  $C_1$ - $C_4$ -haloalkyl,  $C_1$ - $C_4$ -alkoxy and/or  $C_1$ - $C_4$ -alkyl-thio;

- Y sulfur, oxygen or a single bond;
- z sulfur, oxygen,  $-SO_-$ ,  $-SO_2$  or a single bond.

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The compounds of the present invention provide a novel therapeutic potential for the treatment of hypertension, pulmonary hypertension, myocardial infarct, angina pectoris, acute kidney failure, renal insufficiency, cerebral vasospasms, cerebral ischemia, subarachnoid hemorrhages, migraine, asthma, atherosclerosis, endotoxic shock, endotoxin-induced organ failure, intravascular coagulation, restenosis after angioplasty, benign prostate hyperplasia, or hypertension or kidney failure caused by ischemia or intoxication.

30

The good effect of the compounds can be shown in the following tests:

Receptor binding studies

35

Cloned human  $ET_A$  receptor-expressing CHO cells and guinea pig cerebellar membranes with > 60 %  $ET_B$  compared with  $ET_A$  receptors were used for binding studies.

# **40** Membrane preparation

The ET<sub>A</sub> receptor-expressing CHO cells were grown in  $F_{12}$  medium containing 10 % fetal calf serum, 1 % glutamine, 100 U/ml penicillin and 0.2 % streptomycin (Gibco BRL, Gaithersburg, MD, USA).

45 After 48 h, the cells were washed with PBS and incubated with 0.05 % trypsin-containing PBS for 5 min. Neutralization was then carried out with  $F_{12}$  medium, and the cells were collected by cen-

trifugation at 300 x g. To lyze the cells, the pellet was briefly washed with lysis buffer (5 mM Tris-HCl, pH 7.4 with 10 % glycerol) and then incubated at a concentration of 10<sup>7</sup> cells/ml of lysis buffer at 4°C for 30 min. The membranes were centrifuged at 5 20,000 x g for 10 min, and the pellet was stored in liquid nitrogen.

Guinea pig cerebella were homogenized in a Potter-Elvejhem homogenizer and [lacuna] obtained by differential centrifugation 10 at 1000 x g for 10 min and repeated centrifugation of the supernatant at 20,000 x g for 10 min.

Binding assays

15 For the ET<sub>A</sub> and ET<sub>B</sub> receptor binding assay, the membranes were suspended in incubation buffer (50 mM Tris-HCl, pH 7.4 with 5 mM MnCl<sub>2</sub>, 40 μg/ml bacitracin and 0.2 % BSA) at a concentration of 50 μg of protein per assay mixture and incubated with 25 pM [125I]-ET<sub>1</sub> [sic] (ET<sub>A</sub> receptor assay) or 25 pM [125I]-RZ<sub>3</sub> [sic]
20 (ET<sub>B</sub> receptor assay) in the presence and absence of test substance at 25°C. The nonspecific binding was determined using 10<sup>-7</sup> M ET<sub>1</sub>. After 30 min, the free and bound radioligand were separated by filtration through GF/B glass fiber filters (Whatman, England) on a Skatron cell collector (Skatron, Lier, Norway) and the filters were washed with ice-cold Tris-HCl buffer, pH 7.4 with 0.2 % BSA. The radioactivity collected on the filters was quantified using a Packard 2200 CA liquid scintillation counter.

Functional in vitro assay system to look for endothelin receptor 30 (subtype A) antagonists

This assay system is a functional, cell-based assay for endothelin receptors. When certain cells are stimulated with endothelin 1 (ET1) they show an increase in the intracellular calcium concentration. This increase can be measured in intact cells loaded with calcium-sensitive dyes.

1-Fibroblasts which had been isolated from rats and in which an endogenous endothelin receptor of the A subtype had been detected 40 were loaded with the fluorescent dye Fura 2-an as follows: after trypsinization, the cells were resuspended in buffer A (120 mM NaCl, 5 mM KCl, 1.5 mM MgCl<sub>2</sub>, 1 mM CaCl<sub>2</sub>, 25 mM HEPES, 10 mM glucose, pH 7.4) to a density of 2 x 10<sup>6</sup>/ml and incubated with Fura 2-am (2 μM), Pluronics F-127 (0.04 %) und DMSO (0.2 %) at 45 37°C in the dark for 30 min. The cells were then washed twice with buffer A and resuspended at 2 x 10<sup>6</sup>/ml.

The fluorescence signal from 2 x 10<sup>5</sup> cells per ml with Ex/Em 380/510 was recorded continuously at 30°C. To the cells were the test substances and, after an incubation time of 3 min, ETl was the maximum change in the fluorescence determined. [sic] The 5 response of the cells to ETl without previous addition of a test substance was used as control and was set equal to 100 %.

Testing of ET antagonists in vivo

10 Male SD rats weighting 250-300 g were anesthetized with amobarbital, artifically ventilated, vagotomized and pithed. The carotid artery and jugular vein were catheterized.

In control animals, intravenous administration of 1  $\mu$ g/kg ET1 15 leads to a distinct rise in blood pressure which persists for a lengthy period.

The test animals received an i.v. injection of the test compounds (1 ml/kg) 5 min before the administration of ET1. To determine 20 the ET-antagonistic properties, the rise in blood pressure in the test animals was compared with that in the control animals.

Endothelin-1-induced sudden death in mice

tended, where appropriate up to several hours.

25 The principle of the test is the inhibition of the sudden heart death caused in mice by endothelin, which is probably induced by constriction of the coronary vessels, by pretreatment with endothelin receptor antagonists. Intravenous injection of 10 nmol/kg endothelin in a volume of 5 ml/kg of body weight results in death 30 of the animals within a few minutes.

The lethal endothelin-1 dose is checked in each case on a small group of animals. If the test substance is administered intravenously, the endothelin-1 injection which was lethal in the reference group usually takes place 5 min thereafter. With other modes of administration, the times before administration are ex-

The survival rate is recorded, and effective doses which protect 40 50 % of the animals (ED 50) from endothelin-induced heart death for 24 h or longer are determined.

Functional test on vessels for endothelin receptor antagonists

Segments of rabbit aorta are, after an initial tension of 2 g and a relaxation time of 1 h in Krebs-Henseleit solution at  $37^{\circ}$ C and pH 7.3-7.4, first induced to contract with K<sup>+</sup>. After washing out, an endothelin dose-effect plot up to the maximum is constructed.

5

Potential endothelin antagonists are administered to other preparations of the same vessel 15 min before starting the endothelin dose-effect plot. The effects of the endothelin are calibrated as a % of the K+-induced contraction. Effective endothelin antagon-ists result in a shift to the right in the endothelin dose-effect plot.

The compounds according to the invention can be administered orally or parenterally (subcutaneously, intravenously, intra15 muscularly, intraperotoneally [sic]) in a conventional way.

Administration can also take place with vapors or sprays through the nasopharyngeal space.

The dosage depends on the age, condition and weight of the 20 patient and on the mode of administration. The daily dose of active substance is, as a rule, about 0.5-50 mg/kg of body weight on oral administration and about 0.1-10 mg/kg of body weight on parenteral administration.

25 The novel compounds can be used in conventional solid or liquid pharmaceutical forms, eg. as uncoated or (film-)coated tablets, capsules, powders, granules, suppositories, solutions, ointments, creams or sprays. These are produced in a conventional way. The active substances can for this purpose be processed with conventional pharmaceutical aids such as tablet binders, fillers, preservatives, tablet disintegrants, flow regulators, plasticizers, wetting agents, dispersants, emulsifiers, solvents, releaseslowing agents, antioxidants and/or propellent gases (cf. H. Sucker et al.: Pharmazeutische Technologie, Thieme-Verlag, Stuttgart, 1991). The administration forms obtained in this way normally contain from 0.1 to 90 % by weight of the active substance.

Synthesis examples

40

Example 1
Methyl 2-hydroxy-3-methoxy-3,3-diphenylpropionate

5 g (19.6 mmol) of methyl 3,3-diphenyl-2,3-epoxypropionate were 45 dissolved in 50 ml of absolute methanol and, at 0°C, 0.1 ml of boron trifluoride etherate was added. The mixture was stirred at 0°C for 2 h and at room temperature for a further 12 h. The sol-

vent was distilled out, the residue was taken up in ethyl acetate, washed with sodium bicarbonate solution and water and dried over magnesium sulfate. After removal of the solvent by distillation there remained 5.5 q (88 %) of a pale yellow oil.

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Example 2 Methyl 2-hydroxy-3-phenoxy-3,3-diphenylpropionate

5 g (19.6 mmol) of methyl 3,3-diphenyl-2,3-epoxypropionate and 10 5.6 g (60 mmol) of phenol were heated together at 100°C for 6 h. Removal of the excess phenol by distillation under high vacuum and purification of the residue by chromatography on silica gel with hexane/ethyl acetate mixtures resulted in 4.9 g (77 %) of a pale yellow oil.

15

Example 3 Methyl 2-(4,6-dimethoxy-pyrimidin-2-yloxy)-3-methoxy-3,3-diphenylpropionate

- 20 2.86 g (10 mmol) of methyl 2-hydroxy-3-methoxy-3,3-diphenylpropionate were dissolved in 40 ml of dimethylformamide, and 0.3 q (12 mmol) of sodium hydride was added. The mixture was stirred for 1 h and then 2.2 g (10 mmol) of 4,6-dimethoxy-2-methylsulfonylpyrimidine were added. After stirring at room tempera-25 ture for 24 h, cautious hydrolysis was carried out with 10 ml of water, the pH was adjusted to 5 with acetic acid, and the solvent was removed by distillation under high vacuum. The residue was taken up in 100 ml of ethyl acetate, washed with water and dried over magnesium sulfate, and the solvent was distilled out. The 30 residue was mixed with 10 ml of ether, and the resulting precipitate was filtered off with suction. After drying, 3.48 g (82 %) of a white powder remained. Melting point 81°C.
- 35 Example 4 2-(4,6-Dimethoxy-pyrimidin-2-yloxy)-3-methoxy-3,3-diphenylpropionic acid
- 2.12 g (5 mmol) of methyl 2-(4,6-dimethoxy-pyrimidin-2-yl-40 oxy)-3-methoxy-3,3-diphenylpropionate were dissolved in 50 ml of dioxane, 10 ml of 1 N KOH solution were added, and the mixture was stirred at 100°C for 3 h. The solution was diluted with 300 ml of water and extracted with ethyl acetate to remove unreacted ester. The aqueous phase was then adjusted to pH 1-2 with dilute 45 hydrochloric acid and extracted with ethyl acetate. After drying over magnesium sulfate and removal of the solvent by distillation, the residue was mixed with an ether/hexane mixture, and the

precipitate which formed was filtered off with suction. After drying, 1.85 g (90 %) of a white powder remained. Melting point  $167^{\circ}\text{C}$ 

- 5 Example 5
  - 2-(4,6-dimethoxy-2-pyrimidinyloxy)-3-methoxy-3,3-diphenyl sodium propionate [sic]
- 1.68 g (4 mmol) of 2-(4,6-dimethoxy-2-pyrimidinyloxy)-3-methoxy-10 3,3-diphenylpropionic acid are dissolved in 4 ml of 1N NaOH + 100 ml of water. The solution is freeze-dried, and the sodium salt of the carboxylic acid used is obtained quantitatively.
- 15 Example 6
  - 10 g (34.9 mmol) of methyl 2-hydroxy-3-methoxy-3,3-diphenylpropionate were dissolved in 50 ml each of methanol and glacial acetic acid, 1 ml of  $RuO(OH)_2$  in dioxane was added, and hydrogenation was carried out with  $H_2$  in an autoclave at  $100^{\circ}C$  under
- 20 100 bar for 30 h. The catalyst was filtered off, the mixture was concentrated, mixed with ether and washed with NaCl solution, and the organic phase was dried and concentrated. 10.1 g of methyl 3,3-dicyclohexyl-2-hydroxy-3-methoxypropionate were obtained as an oil.

25

Example 7

Methyl 2-[(4,6-dimethoxypyrimidin-2-yl)thio]-3-methoxy-3,3-diphenylpropionate [sic]

- 30 7.16 g (25 mmol) of methyl 2-hydroxy-3-methoxy-3,3-diphenylpropionate were dissolved in 50 ml of dichloromethane, 3 g (30 mmol) of triethylamine were added, and 3.2 g (28 mmol) of methanesulfonyl chloride were added dropwise while stirring. The mixture was stirred at room temperature for 2 h, washed with water, dried
- 35 over magnesium sulfate and concentrated under reduced pressure. The residue was taken up in DMF and added dropwise at 0°C to a suspension of 12.9 g (75 mmol) of 4,6-dimethoxypyrimidine-2-thiol and 8.4 g (100 mmol) of sodium bicarbonate in 100 ml of DMF. After stirring at room temperature for 2 h and at 60°C for a
- 40 further 2 h, the mixture was poured into 1 liter of ice-water, and the resulting precipitate was filtered off with suction.

  After drying, 3.19 g (29 %) of a white powder remained.

Example 8
Methyl 2-hydroxy-3,3-diphenylbutyrate

1.5 g (5.9 mmol) of methyl 3,3-diphenyl-2,3-epoxypropionate dissolved in 10 ml of absolute ether were added dropwise to a cuprate solution which had been prepared from 635 mg (7 mmol) of copper(I) cyanide dissolved in 10 ml of absolute ether and 8.14 ml (13 mmol) of a 1.6 normal methyllithium solution and had been cooled to -78°C. The solution was stirred at -78°C for 1 h and then allowed to warm to room temperature. It was subsequently diluted with 100 ml of ether and 100 ml of water, and the ether phase was washed with dilute citric acid and with sodium bicarbonate solution and dried over magnesium sulfate. The crude product was purified by chromatography on silica gel with cyclohexane/ethyl acetate mixtures to result in 250 mg (16 %) of a pale yellow oil.

Example 9

2-Hydroxy-3-methoxy-3,3-diphenylpropionic acid

20

91.11 q (0.5 mol) of benzophenone and 45.92 q (0.85 mol) of sodium methoxide were suspended in 150 ml of methyl tert-butyl ether (MTB) at room temperature. After cooling to -10°C, 92.24 g (0.85 mol) of methyl chloroacetate were added in such a way that 25 the internal temperature rose to 40°C while continuing to cool in a bath at  $-10^{\circ}$ C. The mixture was then stirred without cooling at the autogenous temperature for one hour. After addition of 250 ml of water and brief stirring, the aqueous phase was separated off. The MTB phase was washed with 250 ml of dilute sodium chloride 30 solution. After the solvent had been changed to methanol (250 ml), a solution of 1 g of p-toluenesulfonic acid in 10 ml of methanol was added at room temperature. The mixture was stirred at autogenous temperature for one hour and then heated to reflux. While distilling out the methanol, 400 g of a 10 % strength 35 sodium hydroxide solution was added dropwise, and finally 60 ml of water were added. The methanol was distilled out until the bottom temperature reached 97°C. After cooling to 55°C, 190 ml of MTB were added and the mixture was acidified to pH 2 with about 77 ml of concentrated HCl. After cooling to room temperature, the 40 aqueous phase was separated off and the organic phase was concentrated by distilling out 60 ml of MtB [sic]. The product was crystallized by adding 500 ml of heptane and slowly cooling to room temperature. The coarsely crystalline solid was filtered off with suction, washed with heptane and dried to constant weight in a 45 vacuum oven at 40°C.

Yield: 108.9 g (80 %), HPLC > 99.5 % area.

Example 10

S-2-Hydroxy-3-methoxy-3,3-diphenylpropionic acid (racemate resolution with L-proline methyl ester)

- 5 148.8 g of a 30 % strength methanolic sodium methanolate solution (0.826 mol) were added dropwise to 240 g of a 57 % strength methanolic L-proline methyl ester hydrochloride solution (0.826 mol) at room temperature, and 2.4 l of MTB and 225 g (0.826 mol) of 2-hydroxy-3-methoxy-3,3-diphenylpropionic acid 10 were added. After 2680 ml of MTB/methanol mixture had been distilled out with simultaneous dropwise addition of 2.4 l of MTB, the mixture was slowly cooled to room temperature, the crystals (R-2-hydroxy-3-methoxy-3,3-diphenylpropionic acid x L-proline methyl ester) were filtered off with suction, and the solid was 15 washed with 150 ml of MTB. The filtrate was concentrated by distilling out 1.5 l of MTB, and 1.0 l of water was added. The pH was adjusted to 1.2 with concentrated hydrochloric acid at room temperature and, after stirring and phase separation, the aqueous phase was separated off and extracted with 0.4 l of MTB. The com-20 bined organic phases were extracted with 0.4 l of water. The residue after the MTB had been stripped off was dissolved in 650 ml of toluene under reflux, and the product was crystallized by seeding and slow cooling. Filtration with suction, washing with toluene and drying in a vacuum oven resulted in 78.7 g of 25 S-2-hydroxy-3-methoxy-3,3-diphenylpropionic acid (yield 35 % based on the racemate). Chiral HPLC: 100 % pure
- 30 Example 11

HPLC: 99.8 %

S-2-Hydroxy-3-methoxy-3,3-diphenylpropionic acid (racemate resolution with <math>(S)-1-(4-nitrophenyl) ethylamine)

30.5 g (0.184 mol) of (S)-1-(4-nitrophenyl)ethylamine were added
35 to 100 g (0.368 mol) of 2-hydroxy-3-methoxy-3,3-diphenylpropionic acid in 750 ml of acetone and 750 ml of MTB under reflux, the mixture was seeded, boiled under reflux for one hour and slowly cooled to room temperature for crystallization. The crystals (S-2-hydroxy-3-methoxy-3,3-diphenylpropionic acid x (S)-1-(4-ni-trophenyl)ethylamine) were filtered off with suction and washed with MTB. The residue was suspended in 500 ml of water and 350 ml of MTB and then the pH was adjusted to 1.2 with concentrated hydrochloric acid at room temperature, and, after stirring and phase separation, the aqueous phase was separated off and extracted with 150 ml of MTB. The combined organic phases were extracted with 100 ml of water. 370 ml of MTB were distilled out

and then 390 ml of n-heptane were added under reflux, and the

mixture was slowly cooled to room temperature while the product crystallized. Filtration with suction, washing with n-heptane and drying in a vacuum oven resulted in 35.0 g of S-2-hydroxy-3-me-thoxy-3,3-diphenylpropionic acid (yield 35 % based on the race-5 mate).

Chiral HPLC: 100 % pure

HPLC: 99.8 %

Example 12

10 Benzyl 3-methoxy-2-(4-methoxy-6,7-dihydro-5H-cyclopentapyrimi-din-2-yloxy)-3,3-diphenylpropionate

24.48 g (90 mmol) of 3-methoxy-3,3-diphenyl-2-hydroxypropionic acid were dissolved in 150 ml of DMF, and 13.7 g (99 mmol) of 15 potassium carbonate were added. The suspension was stirred at room temperature for 30 min. Then 10.7 ml (90 mmol) of benzyl bromide were added dropwise over the course of 5 min, and the mixture was stirred for 1 h, during which the temperature rose to 32°C.

20

To this mixture were successively added 24.84 g (180 mmol) of  $K_2CO_3$  and 20.52 g (90 mmol) of 2-methanesulfonyl-4-methoxy-6,7-dihydro-5H-cyclopentapyridine [sic], and the mixture was stirred at 80°C for 3 h.

25

For workup, the contents of the flask were diluted with about 600 ml of  $\rm H_2O$  and cautiously acidified with concentrated HCl, and 250 ml of ethyl acetate were added. 31.4 g of pure product precipitated and were filtered off.

30

The ethyl acetate phase was separated from the mother liquor, the aqueous phase was extracted again with ethyl acetate, and the combined organic phases were concentrated. The oily residue (19 g) was purified by chromatography (cyclohexane/ethyl acet35 ate = 9/1) to result in a further 10.5 g of pure product.

Total yield: 41.9 g (82.2 mmol)  $\triangleq$  91 % Melting point 143-147°C MS: MH+ = 511

40

Example 13

3-Methoxy-2-(4-methoxy-(6,7-dihydro-5H-cyclopentapyrimidin-2-yl-oxy)-3,3-diphenylpropionic [sic] acid

45 40 g (78.4 mmol) of benzyl 3-methoxy-2-(4-methoxy-6,7-di-hydro-5H-cyclopentapyrimidin-2-yloxy)-3,3-diphenylpropionate were dissolved in 400 ml of ethyl acetate/methanol (4:1), about 500 mg

of palladium on active carbon (10 %) were added, and the mixture was exposed to a hydrogen atmosphere until no further gas was taken up. The catalyst was filtered off, the solution was evaporated, and the residue was crystallized from ether.

Example 14
Ethyl 2S-3,3-diphenyloxirane-2-carboxylate

2.57 g (10.2 mmol) of ethyl 3,3-diphenylacrylate and 464 mg of
10 4-phenylpyridine N-oxide were dissolved in 24 ml of methylene
chloride, and 432 mg (6.5 mol%) of (5,5)-(+)-N,N'-bis(3,5-ditert-butylsalicylidene)-1,2-cyclohexanediaminomanganese(III)
[sic] chloride were added. While cooling in ice, 6.4 ml of a 12 %
strength sodium hypochloride [sic] solution were added, and the
15 mixture was stirred while cooling in ice for 30 min and at room
temperature overnight. The solution was diluted to 200 ml with
water, extracted with ether, dried and evaporated. 2.85 g of a
colorless oil were obtained. Purification by NPLC [sic] (cyclohexane:ethyl acetate = 9:1) resulted in 1.12 g of oil with an
20 enantiomer ratio of about 8:1 in favor of the S configuration.

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<sup>1</sup>H-NMR [CDCl<sub>3</sub>], \delta = 1.0 (t, 3H); 3.9 (m, 3H); 7.3 (m, 10H)
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**25** Example 15

2-Methylsulfonyl-6,7-dihydro-5H-cyclopentapyrimidin-4-ol [sic]

46.9 g (330 mmol) of methyl cyclopentanone-2-carboxylate and 53.5 g (192 mmol) of 5-methylisothiourea [sic] sulfate were suc30 cessively added to 29.6 g (528 mmol) of KOH in 396 ml of methanol, and the mixture was stirred at room temperature overnight, acidified with 1N hydrochloric acid and diluted with water. The crystals which separated out were filtered off with suction and dried. 20 g of crystals were obtained.

35

Example 16

4-Chloro-2-methylsulfanyl-6,7-dihydro-5H-cyclopentapyrimidine

255 ml of phosphorus oxychloride were added to 20 g (110 mmol) of 40 [lacuna], and the mixture was stirred at 80°C for 3 hours. Phosphorus oxychloride was evaporated off, ice was added to the residue, and the crystals which separated out were filtered off with suction. 18.5 g of a brownish solid were obtained.

Example 17

4-Methoxy-2-methylsulfonyl-6,7-dihydro-5H-cyclopentapyrimidine

18.05 g (90 mmol) of 4-chloro-2-methylthio-6,7-dihydro-5H- cyclo5 pentapyrimidine were dissolved in 200 ml of methanol. At 45°C,
16.7 g of sodium methoxide (as 30 % strength solutions [sic] in
methanol) were added dropwise, and the mixture was stirred for
2 hours. The solution was evaporated, taken up in ethyl acetate
and acidified with dilute hydrochloric acid, and the ethyl acet10 ate extract was evaporated. 15.5 g of an oil remained.

```
<sup>1</sup>H-NMR [DMSO],

\delta = 2.1 (quintet, 2H); 2.5 (s, 3H);

2.8 (dt, 4H); 3.9 (s, 3H) ppm
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15

Example 18

2-Methylsulfonyl-4-methoxy-6,7-dihydro-5H-cyclopentopyrimidine [sic]

20 15 g (76.2 mmol) of 4-methoxy-2-methylsulfonyl-6,7-dihydro-5H-cyclopentapyrimidine [sic] were dissolved in 160 ml of glacial acetic acid/methylene chloride (1:1), and 1.3 g of sodium tungstate were added. At 35°C, 17.5 ml (170 ml [sic]) of a 30 % strength H<sub>2</sub>O<sub>2</sub> solution were added dropwise. The mixture was then 25 diluted with 500 ml of water and 100 ml of methylene chloride, and the organic phase was separated off, dried and evaporated. 14 g of oil remained and were crystallized from ether.

```
<sup>1</sup>H-NMR [CDCl<sub>3</sub>],

30 \delta = 2.2 (quintet, 2H); 3.0 (dt., 4H);

3.3 (s, 3H); 4.1 (s, 3H) ppm
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Example 19

1-Benzenesulfonyl-3-(4,6-dimethoxy-2-pyrimidinyloxy)-4-methoxy-35 4,4-diphenyl-2-butanone

- 0.37 g (2.4 mmol) of phenyl methane [sic] sulfone were dissolved in 10 ml of dry THF and then, at  $-70^{\circ}$ C, 2 eq. of butyllithium (2.94 ml; 1.6 molar solution in hexane) were added dropwise.
- 40 After 1 h at -70°C, 1 g (2.4 mmol) of methyl 2-(4,6-dime-thoxy-2-pyrimidinyloxy)-3-methoxy-3,3-diphenylpropynoate [sic] dissolved in 5 ml of THF was added dropwise. The reaction mixture was then stirred at -70°C for 1 h and at -10°C for 1 h and then warmed to room temperature.

For workup, about 10 ml of saturated NH<sub>4</sub>Cl solution were added dropwise, thorough extraction with ethyl acetate was carried out, and the combined organic phases [lacuna] with saturated N-Cl [sic] solution and dried over Na<sub>2</sub>SO<sub>4</sub>. The residue obtained after 5 drying and concentration was purified by chromatography on silica gel (n-heptane/ethyl acetate 15 % → 30 %) and subsequently MPLC on RP silica gel (acetonitrile/H<sub>2</sub>O + TFA); 0.3 g of a white amorphous powder was obtained as product.

# 10 Example 20

3,3-Diphenyloxiram-2-carbonitrile [sic]

3.1 g (54.9 mmol) of sodium methoxide were suspended in 20 ml of dry THF and then, at -10°C, a mixture of 5 g (27.4 mmol) of benzo15 phenone and 4.2 g (54.9 mmol) of chloroacetonitrile was added dropwise.

The reaction mixture was stirred at  $-10^{\circ}$ C for about 2 h, then poured into water and extracted several times with ethyl acetate.

20 The combined organic phases were dried over  $Na_2SO_4$  and concentrated, and the residue was purified by chromatography on silicatel (n-heptane/ethyl acetate).

Yield: 1.2 g (20 %)

25 <sup>1</sup>H-NMR [CDCl<sub>3</sub>],  $\delta = 3.9$  (s, 1H); 7.4-7.5 (m, 10 H) ppm

Example 21

2-Hydroxy-3-methoxy-3,3-diphenylpropionitrile

30

6.5 [lacuna] (29.4 mmol) of 3,3-diphenyloxirane-2-carbonitrile were dissolved in 60 ml of methanol and, at 0 $^{\circ}$ C, about 2 ml of boron trifluoride etherate solution were added. The mixture was stirred further at 0 $^{\circ}$ C for 1 h and then at room temperature over-

35 night. For workup it was diluted with diethyl ether and washed with saturated NaCl solution, and the organic phase was dried over  $Na_2SO_4$  and concentrated. The residue comprised 7.3 g of a white amorphous powder which was used directly in the subsequent reactions.

40

<sup>1</sup>H-NMR [CDCl<sub>3</sub>],  $\delta = 2.95$  (broad s, OH), 3.15 (s, 3H), 5.3 (s, 1H), 7.3-7.5 (m, 10) ppm Example 22 2-(4,6-Dimethoxy-2-pyrimidinyloxy)-3-methoxy-3,3-diphenylpro-pionitrile

- 5 7.3 g (28.8 mmol) of 2-hydroxy-3-methoxy-3,3-diphenylpropionitrile were dissolved in 90 ml of DMF, and 4 g (28.8 mmol) of K<sub>2</sub>CO<sub>3</sub> and 6.3 g (28 mmol) of 2-methanesulfonyl-4,6-dimethoxypy-rimidine were added. The mixture was stirred at room temperature for about 12 h, then poured into water and extracted with ethyl acetate. The combined organic phases were washed again with H<sub>2</sub>O,
- 10 acetate. The combined organic phases were washed again with  $H_2O$ , dried and concentrated. The residue obtained in this way was then purified by chromatography on silica gel (n-hepane/ethyl acetate).

Yield: 6.9 g of white amorphous powder

15

```
FAB-MS: 392 (M+H+)
^{1}H-NMR [CDCl<sub>3</sub>],
\delta = 3.3 (s, 3H); 4.95 (s, 6H), 5.85 (s, 1H);
6.3 (s, 1H); 7.3-7.5 (m, 10H) ppm
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20

Example 23
5-[2-(4,6-Dimethoxy-2-pyrimidinyloxy)-3-methoxy-3,3-diphenyl)propyl]-1H-tetrazole [sic]

- 25 0.5 g (1.3 mmol) of nitrile was dissolved in 10 ml of toluene, and 85 mg (1.3 mmol) of  $NaN_3$  and 460 mg (1.4 mmol) of  $Bu_3SnCl$  were successively added, and then the mixture was refluxed for about 40 h. Cooling was followed by dilution with ethyl acetate and washing with 10 % aqueous KF solution and with NaCl solution.
- 30 After drying over  $MgSO_4$  and concentration there remained 1.0 g of a yellow oil, which was purified by chromatography on silica gel (n-heptane/ethyl acetate).

Concentration of the fractions resulted in 60 mg of the 1H-tetra-35 zole and 110 mg of the 1-methyltetrazole, each as amorphous white solids.

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5-[2-(4,6-Dimethoxy-2-pyrimidinyloxy)-3-methoxy-3,3-diphenyl)-propyl]-1H-tetrazole [sic]
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40

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Electrospray-MS: 435 (M+H+) ^{1}H-NMR (CDCl<sub>3</sub>): ^{3} (ppm) 3.28 (s, 3H), 3.85 (s, 6H), 5.75 (s, 1H), 7.25-7.40 (m, 10H), 7.50 (s, 1H).
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5-[2-(4,6-Dimethoxy-2-pyrimidinyloxy)-3-methoxy-3,3-diphenyl)-propyl]-1-methyltetrazole [sic]

Electrospray-MS; 471 (M+H+)

5  $^{1}H-NMR$  (CDCl<sub>3</sub>):

 $\delta$  (ppm) 3.0 (s, 3H), 3.35 (s, 3H9 [sic], 3.80 (s, 6H), 5.75 (s, 1H), 7.30-7.40 (m, 11H).

Example 24

10 2-(4,6-Dimethoxy-2-pyrimidinyloxy)-3-methylsulfinyl-3,3diphenylpropionic acid

1.2 g (2.9 mmol) of 2-(4,6-dimethoxy-2-pyrimidinyloxy)-3-methylsulfonyl-3,3-diphenylpropionic [sic] acid were introduced 15 into 15 ml of glacial acetic acid at 0°C and 294  $\mu$ l of 30 % strength  $H_2O_2$  were added dropwise. The mixture was stirred at room temperature overnight, poured into water, extracted with  $CH_2Cl_2$  and washed with sodium thiosulfate solution and brine. After drying, 1 g of substance was isolated as a white foam.

Example 25

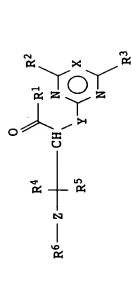
20

2-(4,6-Dimethoxy-2-pyrimidinyloxy)-3-methylsulfonyl-3,3-diphenylpropionic acid

25 0.6 g (1.45 mmol) of 2-(4,6-dimethoxy-2-pyrimidinyloxy)-3-methyl-sulfonyl-3,3-diphenylpropionic [sic] acid was introduced into 15 ml of glacial acetic acid at room temperature, and 294 μl of 30 % strength H<sub>2</sub>O<sub>2</sub> were added dropwise. The mixture was stirred at room temperature overnight, heated at 50°C for a further 3 h, poured into water and washed with sodium thiosulfate solution and brine. After drying, 400 mg were isolated as a white solid.

The compounds listed in Table 1 [sic] can be prepared in a similar way.

35



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No.	R1	R <sup>4</sup> , R <sup>5</sup>	R6	R <sup>2</sup>	R³	Х	λZ		m.p.[°C]
1-1	OMe	Phenyl	Methyl	OMe	ОМе	СН	0 0		81
1-2	НО	Phenyl	Methyl	OMe	ОМе	СН	0 0		167
1-3	но	Phenyl	CH <sub>2</sub> -CH <sub>2</sub> -S-CH <sub>3</sub>	ОМе	ОМе	СН	0 0		
1-4	НО	Phenyl	Ethyl	OMe	ОМе	СН	0 0		81 (decomp.)
1-5	ЮН	Phenyl	iso-Propyl	OMe	ОМе	СН	) (	0	182
9 <u>-</u> I	НО	Phenyl	Methyl	ОМе	ОМе	СН	0   8		168
1-1	НО	Phenyl	CH <sub>2</sub> -CH <sub>2</sub> -SO <sub>2</sub> -CH(CH <sub>3</sub> ) <sub>2</sub>	ОМе	OMe	СН	) (	0	
8-1	НО	Phenyi	CH <sub>2</sub> -CH <sub>2</sub> -SO <sub>2</sub> -CH(CH <sub>3</sub> ) <sub>2</sub>	ОМе	OMe	СН	s (c	0	
6-1	НО	Phenyl	CH <sub>2</sub> -CH <sub>2</sub> -SO <sub>2</sub> -CH(CH <sub>3</sub> ) <sub>2</sub>	OMe	OMe	C-CH(CH <sub>3</sub> ) <sub>2</sub>	0 0	)	
1-10	НО	Phenyl	CH <sub>2</sub> -CH <sub>2</sub> -SO <sub>2</sub> -CH(CH <sub>3</sub> ) <sub>2</sub>	ОМе	OMe	C-CH(CH <sub>3</sub> ) <sub>3</sub>	0 0		
1-11	НО	Phenyl	CH <sub>2</sub> -CH <sub>2</sub> -SO <sub>2</sub> -CH(CH <sub>3</sub> ) <sub>2</sub>	ОМе	NH-OCH3	СН	0 0	)	
1-12	НО	Phenyl	n-Propyi	ОМе	ОМе	СН	0 0		174
I-13	ОМе	Phenyl	n-Propyl	ОМе	ОМе	СН	0 0		
I-14	но	Phenyl	n-Propy1	OEt	0Et	СН	0 0	)	



m.p.[°C] 115 136 173 0-CH(CH<sub>3</sub>)-CH<sub>2</sub>-C 0-CH<sub>2</sub>-CH<sub>2</sub>-C H CH 뚱 CH CH H B B H 뜅 CH 핑 핑 СН H 병 E 몽 CH CH OMe OE OMe OMe OMe ОМе OMe OEt Cyclopropylmethylene [sic] (CH<sub>3</sub>)<sub>2</sub>CH-CH<sub>2</sub>-CH<sub>2</sub> (CH<sub>3</sub>)<sub>3</sub>C-CH<sub>2</sub>-CH<sub>2</sub> 4-Isopropyl-Phenyl 4-Me-O-Phenyl 4-Me-S-Pheny HO-CH<sub>2</sub>-CH<sub>2</sub> HO<sub>2</sub>C-(CH<sub>2</sub>)<sub>2</sub>-2-Me-Phenyl 3-Et-Phenyl Cyclopropyl Cyclopentyl Cyclohexyl tert.-Butyl iso-Butyl iso-Butyl n-Butyl Phenyl Phenyl Methyl Phenyl  $R^6$ Phenyl OMe но HO Ю ЮН но HO ᆼ НО ЮН НО НО Ю I-36 1-24 1-26 1-35 **I-16** I-19 I-23 1-25 **I-27** I-28 I-29 I-30 1-32 1-33 1-34 I-18 1-20 I-17 I-22 **I-31** 1-21



																				$\neg$		
m.p.[ <sup>o</sup> C]																						
Z	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ĬY	0	0	0	S	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
X	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	о-сн=сн-с	СН	СН	СН	СН	СН	СН
$\mathbb{R}^3$	OMe	OMe	эмо	ОМе	OMe	ОМе	ОМе	эМΟ	οМο	эмо	эмо	эмо	эмо	ОМе	эмо	)-0	эмо	эмо	оМе	эмо	0Et	OMe
R <sup>2</sup>	OMe	OMe	OMe	OMe	OMe	OMe	ОМе	OMe	ОМе	ОМе	ОМе	OMe	OMe	ОМе	OMe	ОМе	OMe	OMe	OMe	OMe	OEt	ОМе
R6	2-Cl-Phenyl	3-Br-Phenyl	4-F-Phenyl	4-F-Phenyl	4-CH <sub>3</sub> -Phenyl	3-NO <sub>2</sub> -Phenyl	2-HO-Phenyl	3,4-Dimethoxyphenyl	3,4-Dioxymethylenephenyl [sic]	3,4,5-Trimethoxyphenyl	Benzyl	2-CI-Benzyl	3-Br-Benzyl	4-F-Benzyi	2-Me-Benzyl	2-Me-Benzyl	3-Et-Benzyl	4-iso-Propyl-Benzyl	4-NO <sub>2</sub> -Propyl-Benzyl	2-Me-5-Propyl-Benzyl	2-Me-5-Propyl-Benzyl	4-Me-2-Propyl-Benzyl
R <sup>4</sup> , R <sup>5</sup>	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl	Phenyl
R1	НО	НО	НО	НО	НО	НО	НО	НО	НО	НО	НО	НО	НО	НО	НО	Ю	НО	НО	НО	НО	НО	НО
No.	1-37	1-38	1-39	I-40	1-41	1-42	I-43	I-44	1–45	I-46	I-47	I-48	I-49	I-50	1-51	1-52	1-53	I-54	1-55	1-56	1-57	I-58

No.	R¹	R <sup>4</sup> , R <sup>5</sup>	R6	R <sup>2</sup>	$\mathbb{R}^3$	×	ZΙ	m.p.[°C]
1-59	ОН	Phenyl	3,4-Dioxomethylenebenzyl [sic]	эмо	эМО	СН	0 0	
09-1	НО	4-F-Phenyl	Methyl	эмо	эмо	СН	0 0	163-165 (decomp.)
1-61	ОМе	4-F-Phenyl	Methyl	0Et	OEt	СН	0 0	
1-62	. НО	4-CI-Phenyl	Methyl	ОМе	OMe	СН	0 0	
1-63	НО	4-Me-O-Phenyl	Methyl	ОМе	OMe	СН	0 0	i
1-64	НО	4-Me-O-Phenyl	Ethyl	ОМе	OMe	СН	0	
I-65	НО	4-Me-Phenyl	Methyl	OMe	OMe	CH	0 0	
99-I	НО	4-Me-Phenyl	Methyl	OMe	D-0	0-CH <sub>2</sub> -CH <sub>2</sub> -C	0 0	
L-67	НО	3-CF <sub>3</sub> -Phenyl	n-Propyl	OMe	OMe	СН	0 0	
89-I	НО	3-CF <sub>3</sub> -Phenyl	n-Propyl	ОМе	)но-о	0-СН(СН3)-СН2-С	0 0	
69-I	НО	4-NO <sub>2</sub> -Phenyl	Methyl	ОМе	OMe	СН	0 0	
<b>I-</b> 70	НО	4-NO <sub>2</sub> -Phenyl	Methyl	эМо	0-0	о-сн=сн-с	0 0	
1-71	НО	3-CI-Phenyl	Ethyl	оМе	OMe	СН	0 0	
1-72	Ю	2-F-Phenyl	Methyl	ОМе	эмо	СН	0 0	193-194 (decomp.)
I-73	НО	2-F-Phenyl	Methyl	оМе	ОМе	СН	0 8	
I-74	НО	2-Me-O-Phenyl	Methyl	оМе	оМе	СН	0 0	
1-75	НО	2-Me-O-Phenyl	Methyl	оМе	ОМе	СН	s o	
1-76	НО	3,4-Dimethoxyphenyl	Methyl	OMe	ОМе	СН	0 0	
1-11	НО	3,4-Dioxomethylenephenyl [sic]	Methyl	ОМе	ОМе	СН	0 0	
I-78	НО	p-CF <sub>3</sub> -Phenyl	Methyl	ОМе	ОМе	СН	0 0	



126 (decomp.) 157 (decomp.) (decomb.) 149-151 118 0 CH2-CH2-CH2-CH2-C CH2-CH2-CH2-C S-C(CH<sub>3</sub>)=C(CH<sub>3</sub>)-C N(CH<sub>3</sub>)-CH=CH-C O-CH(CH<sub>3</sub>)-CH<sub>2</sub>-C O-C(CH<sub>3</sub>)=CH-C O-C(CH<sub>3</sub>)=CH-C CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-C CH<sub>2</sub>-CH<sub>2</sub>-C O-CH<sub>2</sub>-CH<sub>2</sub>-C 0-CH<sub>2</sub>-CH<sub>2</sub>-C 0-CH=CH-C S-CH=CH-C CH CH 동 НЭ 동 CH CH Z × NH-OMe OCF<sub>3</sub> OMe OEt OEt  $CF_3$  $CF_3$ Me Me Methyl OMe OMe OMe OMe Ethyl OMe Me Me Me n-Propyl Methyl Benzyl Methyl Methyl Methyl Methyl Methyl Methyl Methyl Ethyl Н 3,4-Dimethoxyphenyl 3,4-Dimethoxyphenyl p-Me-O-Phenyl 4-F-phenyl 4-F-phenyl Phenyl Phenyl Phenyl  $R^4, R^5$ Phenyl OMe НО НО HO НО HO НО HO HO НО Ю ЮН Ю НО ОН НО НО НО НО НО но HO ~ I-100 98-I I-93 I-94 I-95 I-99 I-82 I-83 I-84 I-85 I-92 96<del>-</del>I I-80 I-81 I-87 **I-88** 68-I **1**-90 I-97 86-I **I-91** 



_	_	_	_	_	_					_	_			_		_	_			_	_	_
m.p.[°C]							134															
Z	0	0	0	0	0	S	S	S	0	0	0	0	0	S	0	0	0	0	0	S	0	0
<u> </u>	0	0	0	0	0	S	0	S	0	0	0	S	0	0	0	S	0	0	0	0	0	0
X	СН	с-сн3	CH	СН	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	H	H	CH	HO	H	z	CH	СН	СН	СН	СН	СН	СН	СН	N	N	Z
$\mathbb{R}^3$	亞	Me	Me	OMe	CH <sub>2</sub> -(	ОСН3	OCH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	ОСН3	ОСН3	$CF_3$	CF <sub>3</sub>	$0$ CF $_3$	СН3	CI	ОСН3	0СН3	0СН3	$0$ CF $_3$	СН3	CI
R <sup>2</sup>	茁	Me	OMe	ОМе	ОМе	0СН3	0СН3	ОСН3	OCH <sub>3</sub>	0СН3	0СН3	$\mathrm{CF}_3$	OCH <sub>3</sub>	0СН3	оснз	0СН3	0СН3	0СН3	0СН3	0СН3	0СН3	0СН3
R6	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Ethyl	Propyl	i-Propyl	s-Butyl	Methyl	Methyl	Methyl	1-Phenylpropyn-3-yl	Methyl	Methyl
R <sup>4</sup> , R <sup>5</sup>	Phenyl	Phenyl	Phenyl	Cyclohexyl	Cyclohexyl	Phenyl	Phenyl	Phenyl	Phenyl	2-Fluorophenyl	3-Chlorophenyi	4-Bromophenyl	Phenyl	Phenyl	Phenyl	Phenyl	2-Methylphenyl	3-Methoxyphenyl	4-Nitrophenyl	Phenyl	2-Hydroxyphenyl	3-Trifluoromethylphenyl
R¹	НО	НО	НО	НО	НО	но	НО	0СН3	НО	ОСН3	OC <sub>2</sub> H <sub>5</sub>	ON(CH <sub>3</sub> ) <sub>2</sub>	0-СН2-С=СН	НО	OCH <sub>3</sub>	OC <sub>2</sub> H <sub>5</sub>	ON(CH <sub>3</sub> ) <sub>2</sub>	ON(CH <sub>3</sub> ) <sub>2</sub>	ON=C(CH <sub>3</sub> ) <sub>2</sub>	ON(CH <sub>3</sub> ) <sub>2</sub>	ON=C(CH <sub>3</sub> ) <sub>2</sub>	ONSO <sub>2</sub> C <sub>6</sub> H <sub>5</sub>
No.	1-101	1-102	1-103	I-104	I-105	<b>I-</b> 106	1-107	I-108	I-109	I-110	1-111	1-112	1-113	1-114	1-115	I-116	1-117	I-118	I-119	I-120	1-121	1-122

m.p.[°C] 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 S 0 0 0 0 0 0 0 0 0 0 0 0 0 S 0 0 0 S S CH CH CH CH H CH H H lΞ CH 핑 H CH Z -0-CH2-CH2--0-CH<sub>2</sub>-CH<sub>2</sub>-핑 뚱 H ОСН3 OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub>  $0CH_3$ OCH<sub>3</sub> OCH<sub>3</sub> OCF3  $CH_3$ CH<sub>3</sub>  $CF_3$  $CF_3$  $\mathbb{R}^3$ ひ  $\Box$ OCH<sub>3</sub> OCH<sub>3</sub> OCH<sub>3</sub>  $0CH_3$ OCH<sub>3</sub> OCH<sub>3</sub>  $0CH_3$ OCH<sub>3</sub> OCH<sub>3</sub>  $0CH_3$ Phenyl  $0CH_3$ OCH<sub>3</sub> OCH<sub>3</sub>  $0CH_3$ OCH<sub>3</sub> OCH<sub>3</sub> CH<sub>3</sub> ರ 2-Methoxyethyl 4-Bromophenyl 2-Fluorophenyl 3-Chloropheny Trifluoroethy 2-Thiazolyl 4-Thiazolyl Phenyl Methyl Benzyl Phenyl  $\mathbb{R}^6$ 4-Dimethylaminophenyl 2-Methylphenyl 2-Fluorophenyl 3-Chlorophenyl 4-Bromophenyl -(CH<sub>2</sub>)<sub>5</sub>-Phenyl Phenyl 0-CH<sub>2=</sub>CH [sic] ON(CH<sub>3</sub>)<sub>2</sub> ON(CH<sub>3</sub>)<sub>2</sub> ON(CH<sub>3</sub>)<sub>2</sub> I-144 ON(CH<sub>3</sub>)<sub>2</sub> 1-123 NHPhenyl -124 OC<sub>2</sub>H<sub>5</sub> I-143 OC2H5 I-138 OC2H5 -142 OCH<sub>3</sub> -137 OCH<sub>3</sub> НО НО НО НО но I-136 OH I-131 OH I-132 OH I-133 OH I-134 OH I-135 OH R I-125 I-139 1-140 I-126 1-130 -141 1-127 1-128 1-129



No.	$\mathbb{R}^1$	R <sup>4</sup> , R <sup>5</sup>	R6	R <sup>2</sup>	$\mathbb{R}^3$	X	ZΑ	m.p.[°C]
1-145	ON=C(CH <sub>3</sub> ) <sub>2</sub>	3-Methoxyphenyl	Phenyl	OCH <sub>3</sub>	OCH <sub>3</sub>	СН	0	
1-146	НО	Phenyl	Methyl	OCH <sub>3</sub>	-CH <sub>2</sub> -(	-CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0	
I-147	НО	4-Fluorophenyl	Methyl	ОСН <sub>3</sub>	6Н3	НЭ	0 0	168 (decomp.)
I-148	НО	4-Fluorophenyl	Methyl	ОСН <sub>3</sub>	-CH <sub>2</sub> -(	-CH <sub>2</sub> -CH <sub>2</sub> -CH	0	
I-149	NH-SO-C <sub>6</sub> H <sub>5</sub>	4-Nitrophenyl	Pheny!	OCH <sub>3</sub>	OCH <sub>3</sub>	CH	0	
I-150	ОСН <sub>3</sub>	Phenyl	3-Imidazolyl	OCH <sub>3</sub>	)-0-	O-CH <sub>2</sub> -CH <sub>2</sub>	0	
I-151	OC <sub>2</sub> H <sub>5</sub>	Phenyl	4-Imidazolyl	OCH <sub>3</sub>	$\mathrm{CF}_3$	z	s 0	
I-152	ON(CH <sub>3</sub> ) <sub>2</sub>	Phenyl	2-Pyrazolyl	OCH <sub>3</sub>	OCF <sub>3</sub>	Z	S 0	
1-153	ON=C(CH <sub>3</sub> ) <sub>2</sub>	2-Hydroxyphenyl	Phenyl	OCH <sub>3</sub>	$\mathrm{CH}_3$	Z	0 0	
I-154	NH-SO <sub>2</sub> -C <sub>6</sub> H <sub>5</sub>	3-Trifluoromethylphenyl	Phenyl	ОСН3	CI	z	0	
1-155	NHPhenyl	4-Dimethylaminophenyl	Phenyl	$0$ CH $_3$	осн3	НЭ	0 S	
I-156	ONa	Phenyl	Phenyl	OCH <sub>3</sub>	OCH <sub>3</sub>	НЭ	S S	
1-157	0-CH <sub>2</sub> -C≡C	Phenyl	Phenyl	OCH <sub>3</sub>	OCH <sub>3</sub>	Z	S S	
I-158	НО	Phenyl	Phenyl	CF <sub>3</sub>	$\mathrm{CF}_3$	НЭ	s o	
I-159	0СН3	Phenyl	Phenyl	OCF3	$0CF_3$	НЭ	0 0	
I-160	OC <sub>2</sub> H <sub>5</sub>	Phenyl	2-Dimethylaminophenyl	CH <sub>3</sub>	$^{ m cH}^{ m 3}$	НЭ	0 0	
1-161	ON(CH <sub>3</sub> ) <sub>2</sub>	Phenyl	3-Hydroxyphenyl	CI	IJ	НЭ	0 0	
1-162	ON=C(CH <sub>3</sub> ) <sub>2</sub>	Phenyl	4-Trifluoromethylphenyl	0СН3	)-0-	O-CH <sub>2</sub> -CH <sub>2</sub> -	s 0	
1-163	NH-SO <sub>2</sub> -C <sub>6</sub> H <sub>5</sub>	Phenyl	2-Oxazolyl	0СН3	$\mathrm{CF}_3$	N	SS	
1-164	НО	Phenyl	Methyl	$CH_3$	$^{ m cH}^{ m 3}$	НЭ	0 0	
1-165	НО	Cyclohexyl	Methyl	ОСН3	<sup>6</sup> НЭО	НЭ	0 0	
1-166 ОН	НО	Cyclohexyl	Methyl	ОСН3	CH <sub>2</sub> -(	СН2-СН2-СН-С	0 0	



No.	R1	R <sup>4</sup> , R <sup>5</sup>	R6	$\mathbb{R}^2$	R <sup>3</sup>	×	Zλ	m.p.[°C]
1-167	НО	Phenyl	Methyl	N(CH <sub>3</sub> ) <sub>2</sub>	N(CH <sub>3</sub> ) <sub>2</sub>	СН	0	
I-168	НО	Phenyl	Methyl	ОСН3	0СН3	CH	0 SO2	2
I-169	НО	Phenyl	Methyl	ОСН3	OCH <sub>3</sub>	CH	0 \$0	2
I-170	НО	3-F-Phenyl	Me	ОМе	OMe	HJ	0	
1-171	НО	3-F-Phenyl	Me	ОМе	CH <sub>2</sub> -C	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0	
1-172	ОН	4-F-Phenyl	Ме	ОМе	O-ZHO	CH <sub>2</sub> -CH <sub>2</sub> -C	0 0	142-143 191°C
I-173	ОН	3-MeO-Phenyl	Ме	ОМе	CH <sub>2</sub> -C	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0	158-161 (decomp.)
1-174	0Н	3-MeO-Phenyl	Me	эМО	οМο	НЭ	0 0	
1-175	НО	3-MeO-Phenyl	Et	ОМе	CH <sub>2</sub> -C	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0	
1-176	НО	Phenyl	HO-CH <sub>2</sub> -CH <sub>2</sub>	ОМе	CH <sub>2</sub> -C	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0	
1-177	НО	Phenyl	Me	NMe <sub>2</sub>	NMe <sub>2</sub>	z	0	181
I-178	но	Phenyl	Me	оМе	ОМе	z	0	
I-179	НО							
I-180	NH-SO <sub>2</sub> -Phenyl	Phenyl	Ме	ОМе	ОМе	СН	0	
1-181	NH-SO <sub>2</sub> -Me	Phenyl	Ме	OMe	OMe	СН	0 0	
I-182	CH <sub>2</sub> -SO <sub>2</sub> -Phenyl	Phenyl	Me	ОМе	OMe	СН	0 0	
1-183	CH <sub>2</sub> -SO <sub>2</sub> -Me	Phenyl	Me	ОМе	ОМе	СН	0 0	
I-184	-CN	Phenyl	Me	ОМе	ОМе	СН	0 0	
1-185	Tetrazole [sic]	Phenyl	Ме	ОМе	OMe	СН	0 0	
I-186	NH-SO <sub>2</sub> -Phenyl	Phenyl	Ме	OMe	ОМе	СН	0 0	167

No.	R1	R <sup>4</sup> , R <sup>5</sup>	R6	R <sup>2</sup>	R³	X	Zλ		m.p.[°C]
I-187	I-187 N-Methyltetrazole [sic]	Phenyl	Me	ОМе	ОМе	СН	0	0	
I-188 ONa		Phenyl	Me	ОМе	-0-CF	O-CH <sub>2</sub> -CH <sub>2</sub> -C-	0 0	0	122-139 (decomp.)
I-189 OH	НО	o-F-Phenyl	Me	ОМе	-0-CF	·0-СН <sub>2</sub> -СН <sub>2</sub> -С-	0	0	140-144 (decomp.)
I-190 OH	НО	m-Me-Phenyl	Me	OMe	OMe	СН	0 0	0	169-177
I-191 OH	НО	m-Me-Phenyl	Me	ОМе	-0-CF	.0-СН <sub>2</sub> -СН <sub>2</sub> -С-	0	0	119-135 (decomp.)
І-192 ОН	но	p-F-Phenyl	Ме	ОМе	Ме	СН	0 0	0	137-140 (decomp.)
I-193 OH	НО	m-F-Phenyl	Me	Me	-0-CF	-0-СН <sub>2</sub> -СН <sub>2</sub> -С-	0 0	0	150-152
I-194 OH	НО	p-F-Phenyl	Me	Me	-0-CI	-0-CH <sub>2</sub> -CH <sub>2</sub> -C-	0 0	0	169-170

, R <sup>2</sup>	×	R <sup>3</sup>
	CH — Y —	*
	\ / Z	
	86	

				41				:							
Y Z m.p.[°C]	86-96							0 0 137-139							
Z	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0	0 0
		)	)	)	)	)	)	)	)	)	)		)	)	<u>)</u>
×	CH	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	СН	O-CH <sub>2</sub> -CH <sub>2</sub> -C	0-сн=сн-с
R <sup>3</sup>	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	0-CH	но-0
R <sup>2</sup>	ОМе	ОМе	ОМе	OMe	OMe	ОМе	ОМе	ОМе	ОМе	ОМе	ОМе	OMe	ОМе	ОМе	ОМе
R6	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Methyl	Isopropyl	p-Isopropylphenyl	Benzyl	Ethyl	(CH <sub>3</sub> ) <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub>	Cyclopropylmethylene [sic]	Methyl	Ethyl
A	Bond	CH <sub>2</sub>	CH <sub>2</sub> -CH <sub>2</sub>	СН=СН	0	S	NH(CH <sub>3</sub> )	Bond	Bond	Bond	СН=СН	СН=СН	СН=СН	СН=СН	CH <sub>2</sub> -CH <sub>2</sub>
۳. آ	ОН	НО	НО	ЮН	ЮН	НО	ОН	HO	ОН	ЮН	HO	ЮН	OH	НО	НО
No.	11-1	II-2	II-3	11-4	11-5	9-11	II-7	8-11	6-11	II-10	II-11	II-12	11-13	II-14	11-15

No.	$\mathbb{R}^1$	А	R6	R <sup>2</sup>	R <sup>3</sup> X	Y Z m.p.[ºC]
II-16	НО	CH <sub>2</sub> =CH <sub>2</sub>	Methyl	ОМе	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0 0
II-17	НО	Bond	Methyl	ОМе	CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C	0 0 147

Example 35

Receptor binding data were measured by the binding assay de-5 scribed above for the compounds listed below.

The results are shown in Table 2 [sic].

Table 2 [sic]

10

1 .

Receptor binding data (Ki values)

	Compound	ET <sub>A</sub> [nM]	ET <sub>B</sub> [nM]
15			
	I-2	6	34
	1-29	86	180
	I-5	12	160
20	I-4	7	2500
20	I-87	1	57
	I.89	86	9300
	I-103	0.4	29
	I-107	3	485
25	I-12	19	1700
	I-26	23	2000
	I-23	209	1100
	I-47	150	1500
30	I-60	33	970
	I <b>-</b> 96	0.6	56
	II-3	107	7300
35	II-1	28	2300

Novel carboxylic acid derivatives, their preparation and use

Abstract

Carboxylic acid derivatives

5

1 3

$$\begin{array}{c|c}
R^{4} & N & \\
R - Z - C - CH - Y & N \\
\downarrow & \downarrow \\
R^{5} & R
\end{array}$$

$$\begin{array}{c|c}
R^{2} \\
X \\
R^{3}$$

### 15

where  $R-R^6$ , X, Y and Z have the meanings stated in the description, and the preparation thereof, are described. The novel compounds are suitable for controlling diseases.

#### 20

## 25

# 30

## 35